SYNLETT Spotlight 253

This feature focuses on a reagent chosen by a postgraduate, highlighting the uses and preparation of the reagent in current research

Diphenylphosphoryl Azide (DPPA) – A Reagent with Manifold Applications

Compiled by Huan Liang

Huan Liang was born in Tianjin, P. R. of China in 1980. He received his B.Sc. in Chemistry from Tianjin University in 2003. He is now pursuing his Ph.D. under the supervision of Prof. Marco Ciufolini, Canada Research Chair in Synthetic Organic Chemistry, at the University of British Columbia. His research mainly focuses on natural product synthesis for pharmaceutical application.

Taylor group.9

OMe

ÓМе

MeC

Department of Chemistry, University of British Columbia, Vancouver, BC, V6T 1Z1, Canada E-mail: lianghuan@chem.ubc.ca



96%

B = COOH

R = NHCbz

 $R = NH_2$

This document was downloaded for personal use only. Unauthorized distribution is strictly prohibited.

Introduction

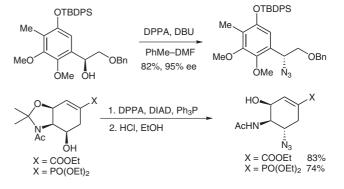
Diphenylphosphoryl azide, originally developed by Yamada in 1972,¹ has shown significant synthetic versatility,² being used in isocyanate synthesis, especially in the Curtius rearrangement,¹ stereospecific conversion of alcohol into azide,³ as a coupling reagent in macrolactamization,⁴ in allylic amine synthesis,⁵ and in aziridination reactions.⁶

Diphenylphosphoryl azide, also called DPPA, diphenyl phosphorazidate or phosphoric acid diphenyl ester azide, is a colorless liquid with high boiling point ($157 \degree C/0.17$ mmHg), and can be easily prepared by the reaction be-

Abstracts

(A) Yamada and co-workers developed an improved method for the Curtius rearrangement reaction using DPPA, which was later named Yamada–Curtius rearrangement.¹ In 2007, the Ciufolini group employed this method in the total synthesis of streptonigrone, to transform a carboxylic acid group into a protected amino group through the hydrolysis of an isocyanate intermediate.¹⁰

(B) A primary or secondary alcohol can be easily converted into an azide group by DPPA under mildly basic conditions or using Mitsunobu conditions for stereochemical inversion. In the total synthesis of cribrostatin VI, the Danishefsky group successfully employed DPPA to displace a benzyl alcohol in high yield and ee.¹¹ Another example was demonstrated in the synthesis of Tamiflu and its phosphonate congeners by the Wang group in 2007.¹²



tween diphenylphosphoryl chloride and sodium azide in acetone in high yield.^{1,7} The Waldvogel group developed

a reliable protocol for the large-scale (100 g) synthesis of

DPPA, including purification by reduced-pressure distil-

lation (Scheme 1).⁸ A polymer-supported form of the re-

agent has also been developed using phenol resin by the

NaN₃ (1.2 equiv), acetone

reduced-pressure distillation

OBn

Me OBn

OMe

ÓMe

DPPA. Et₃N. PhH:

then 1 M LiOH

CbzCl, NaHCO₃

70% over 3 steps

Scheme 1 Preparation of DPPA in 400 mmol scale

SYNLETT 2008, No. 16, pp 2554–2555 Advanced online publication: 02.07.2008 DOI: 10.1055/s-2008-1067132; Art ID: V25908ST © Georg Thieme Verlag Stuttgart · New York (C) Diphenylphosphoryl azide has also been widely used in peptide coupling reactions, particularly in macrolactamization.⁴ In 2005, the Moody group completed the synthesis of thiopeptide amythiamicin D. In the final step, after global deprotection of *N*-Boc and *tert*-butyl groups, an α -amino ketone was successfully coupled with a thiazole carboxylic acid in DMF in 73% yield.¹³

OH

Me

NP(O)(OPh)₂

2. DPPA, PhH

റ്

PhH

Me

1 M HC

MeOH

85% over 2 steps

Me

Ме

PdCl₂(MeCN)₂

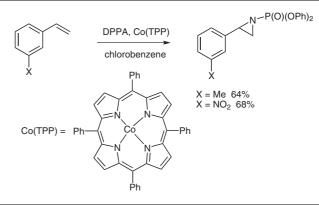
CH₂Cl₂

Me

Me

(D) The Batey group has developed a stereoselective synthesis of allylic amines through a [3,3]-aza-phospha-oxa-Cope sigmatropic rearrangement.⁵ Methylvinylcarbinol was converted into crotylamine in 85% yield over two steps. DPPA was used as an amine source in these reactions, and excellent selectivity was achieved through addition of a catalytic amount of $PdCl_2(MeCN)_2$ catalyst.

(E) A new catalytic aziridination reaction using cobalt tetraphenylporphyrin [Co(TPP)] as catalyst has been extensively studied by the Zhang group.⁶ DPPA functioned as a nitrene source in the reaction that proceeded in good to excellent yield.



References

- (a) Shioiri, T.; Ninomiya, K.; Yamada, S. J. Am. Chem. Soc. 1972, 94, 6203. (b) Ninomiya, K.; Shioiri, T.; Yamada, S. Tetrahedron 1974, 30, 2151.
- (2) Bräse, S.; Gil, C.; Knepper, K.; Zimmermann, V. Angew. Chem. Int. Ed. 2005, 44, 5188.
- (3) (a) Thompson, A. S.; Humphrey, G. R.; DeMarco, A. M.; Mathre, D. J.; Grabowski, E. J. J. J. Org. Chem. 1993, 58, 5886. (b) Hughes, D. L. Org. React. 1992, 42, 335.
- (4) Han, S. Y.; Kim, Y. A. Tetrahedron 2004, 60, 2447.
- (5) (a) Lee, E. E.; Batey, R. A. Angew. Chem. Int. Ed. 2004, 43, 1865; Angew. Chem. 2004, 116, 1901. (b) Lee, E. E.; Batey, R. A. J. Am. Chem. Soc. 2005, 127, 14887.
- (6) Gao, G. Y.; Jones, J. E.; Vyas, R.; Harden, J. D.; Zhang, X. P. J. Org. Chem. 2006, 71, 6655.
- (7) Appropriate safety measures are necessary when using DPPA, a source of toxic azide ion. $(LD_{50} \text{ of azide ion } 29 \text{ mg/} \text{kg in rats, calculated from } LD_{50} \text{ of } NaN_3 \text{ as per Merck index}$). Also, like all organic azides, DPPA should be regarded as potentially explosive.
- (8) Wolff, O.; Waldvogel, S. R. Synthesis 2004, 1303.
- (9) Lu, Y.; Taylor, R. T. Tetrahedron Lett. 2003, 44, 9267.
- (10) Chan, B. K.; Ciufolini, M. A. J. Org. Chem. 2007, 72, 8489.
- (11) Chan, C.; Heid, R.; Zheng, S.; Guo, J.; Zhou, B.; Furuuchi, T.; Danishefsky, S. J. J. Am. Chem. Soc. 2005, 127, 4596.
- (12) Shie, J. J.; Fang, J. M.; Wang, S. Y.; Tsai, K. C.; Cheng, Y. S. E.; Yang, A. S.; Hsiao, S. C.; Su, C. Y.; Wong, C. H. *J. Am. Chem. Soc.* 2007, *129*, 11892.
- (13) Hughes, R. A.; Thompson, S. P.; Alcaraz, L.; Moody, C. J. J. Am. Chem. Soc. 2005, 127, 15644.

This document was downloaded for personal use only. Unauthorized distribution is strictly prohibited.

[∼]Me

NP(O)(OPh)₂

NH₂

·HCI