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Novel Synthesis of 2,3-Bisboryl-1,3-dienes from 1-Bromo-1-lithioethene and 1,1-Bisborylalkenes

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This paper is dedicated to Professor Ryoji Noyori on the occasion of his retirement of Nagoya University, Japan

Abstract: Treatment of 1,1-[bis(pinacolato)boryl]alkenes with excess of 1-bromo-1-lithioethene was found to give 2,3-bis[(pinacolato)boryl]-1,3-dienes in moderate to good yields. Synthetic applications of 2,3-[bis(pinacolato)boryl]-1,3-butadiene were also demonstrated.

Key words: boron, lithium, carbenoids, insertion, alkenes

Alkenylboron compounds are readily accessible and extremely useful reagents in organic synthesis.¹ In contrast, bis(alkenylboron) compounds have attracted less attention probably because their facile syntheses are quite limited,² though bis(alkenylboron) compounds would be employed for an efficient synthesis of polysubstituted olefins through double carbon-carbon bond formation with retention of configuration by a simple experimental operation. Very recently, we found that treatment of bis(pinacolato)diboron 1 or (dimethylphenylsilyl)(pinacolato)boron 3 with 1-halo-1-alkenyllithium gave the corresponding 1,1-bisborylalkenes 2 or 1-silyl-1-borylalkenes 4, respectively.³ In particular, 2 is a parent example of bis(alkenylboron) compounds (eq.).

$$B-M$$

$$B = B(OCMe_2)_2$$

$$1 [M = B(OCMe_2)_2]$$

$$3 [M = SiMe_2Ph]$$

$$M$$

$$B = R'$$

$$B = R'$$

$$B = R'$$

$$A [M = B(OCMe_2)_2]$$

$$A [M = SiMe_2Ph]$$

Equation

During the course of the synthetic studies, we eventually found that 2,3-bisboryl-1,3-dienes **5a** were produced when excess of 1-bromo-1-lithioethene was treated with **1**. Formation of **5a** was ascribed to the reaction of 1,1-bisborylethene **2a** with CH₂=CBrLi followed by 1,2-migration of a carbon (Scheme 1). We report herein that the synthesis of 2,3-bisboryl-1,3-dienes is general, and bisborylated 1,3-dienes **5** serve as useful precursors of complex 1,3-dienes. In addition, introduction of two boryl groups into a 1,3-diene unit enhances the synthetic utility of the addition products.

Scheme 1 $B = B(OCMe_2)_2$

Treatment of 1-bromo-1-lithioethene (1 mol), generated from vinyl bromide and lithium 2,2,6,6-tetramethylpiperidide (LiTMP) in THF-Et₂O (2:1) at -110 °C, with 1,1bis[(pinacolato)boryl]ethene 2a (1 mol) at −110 °C gave 2,3-bisboryl-1,3-butadiene **5a** in 7% yield (Scheme 2). In view that diboron 1 reacts with an equimolar amount of 1bromo-1-lithioethene to give 2a in 91% yield,³ the low yield indicates that the reaction of the carbenoid with 2a is slower than with 1 and apparently competes with the decomposition of the lithium carbenoid. Then, we increased the amount of the carbenoid reagent, and observed that 72% yield was achieved when 5 molar equivalents of vinyl bromide and LiTMP were employed as shown in Scheme 2.^{6,7} Noteworthy is that **5a** can be purified by column chromatography on silica gel since 2-boryl-1,3-diene is reported to be highly susceptible to dimerization.8 Carbenoid generation carried out in the presence of 2a gave **5a** in lower yield (59%), while reaction of 2-substituted 1bromo-1-lithioethene with **2a** did not proceed at all.

$$= \underbrace{\begin{array}{c} LiTMP (n \text{ mol}) \\ THF-Et_2O \\ (2:1) \\ (n \text{ mol}) \end{array}}_{C \text{ mol}} \underbrace{\begin{array}{c} Li \\ Br \\ \end{array}}_{Br} \underbrace{\begin{array}{c} 2a \text{ (1 mol)} \\ -110 \text{ °C to r.t.} \\ \end{array}}_{B}$$

Yield (%) of **5a**: 7 (n = 1), 46 (n = 3), 72 (n = 5), 60 (n = 10)

Scheme 2 Synthesis of 2,3-bisboryl-1,3-diene 5a

The optimized conditions were applied to 2-monosubstituted bisborylethenes **2b** and **2c**. The corresponding conjugated triene **5b** as a E/Z mixture (73:27) and dienyne **5c** as the only E-isomer were respectively isolated in 74% and 38% yields. The stereochemical outcome indicates that 1-bromo-1-lithioethene preferentially attacks the

sterically less hindered boron atom of **2**. Reaction of 2,2-disubstituted-1,1-diborylethenes **2d** and **2e** also took place smoothly, giving rise to **5d** and **5e** in good yields.

LiTMP (5 mol) 2 (1 mol)
$$(5 \text{ mol})$$
 $(2 : 1)$ (5 mol) (5 mol)

Scheme 3 Synthesis of 2,3-bisboryl-1,3-diene 5

One-pot synthesis of **5a** from **1** is possible. As shown in Scheme 4, treatment of vinyl bromide (5 mol) with LiT-MP (5 mol) followed by the addition of diboron **1** produced **5a** in 82% yield.

Scheme 4 One-pot synthesis of 5a from 1

Synthetic utility of 2,3-bisboryl-1,3-butadiene **5a** is demonstrated in Scheme 5. Diels—Alder reaction of **5a** is particularly accelerated by the two boryl groups and indeed proceeded with maleimide even at room temperature to give 1,2-bisborylated cyclohexene **6** in 93% yield.^{8,10} Cross-coupling reaction with iodobenzene catalyzed by Pd(PPh₃)₄ allowed us to prepare 2,3-diphenyl-1,3-butadiene **7**.¹¹ 1,2,3,4-Tetraboryl-2-butene **8** or 1-silyl-2,3,4-tris(boryl)-2-butene **9** was synthesized as a single diastereomer by Pt-catalyzed 1,4-addition reaction with bis(pinacolato)diboron **1** or (dimethylphenylsilyl)(pina-

colato)boron 3.¹² The highly borylated compounds 8 and 9 contain both alkenyl- and allylmetal moieties and thus may serve as valuable synthetic reagents.

toluene, r.t.
$$93\%$$
 Ph —I
 $Pd(PPh_3)_4 (3 \text{ mol}\%)$
 E toluene, 80 °C
 E single diastereomer
 E single diastereomer
 E single diastereomer

Scheme 5 Synthetic applications of 2,3-diboryl-1,3-butadiene

In summary, we have established a novel synthesis of 2,3-bisboryl-1,3-dienes from 1-bromo-1-lithioethene and 1,1-bisborylalkenes. We can transform 2,3-bisboryl-1,3-dienes into various types of complex molecules using the boron functionality as a key element before/after unique reactions of 1,3-dienes. Further studies based on 2,3-bisboryl-1,3-dienes are in progress in our laboratory.

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- (7) Representative procedure of 5a: Butyllithium in hexane (1.56 M, 321 μ L, 0.49 mmol) was added to a solution of 2,2,6,6tetramethylpiperidine (84 μ L, 0.50 mmol) in a mixture of THF (1 mL) and diethyl ether (0.5 mL) at 0 °C, and the resulting solution was stirred at 0 °C for 5 min. To this solution was added a THF solution of vinyl bromide (1.0 M, 500 µL, 0.50 mmol) at -110 °C and 1,1-[bis(4,4,5,5-tetramethyl-1,3,2dioxaborolan-2-yl)]ethene (2a) (25 mg, 0.10 mmol) in THF $(0.1\ \text{mL})$ successively. The resulting mixture was gradually allowed to warm up to room temperature and stirred for 12 h. The reaction mixture was quenched with three drops of sat. aq NH₄Cl, and diluted with diethyl ether (10 mL) and water (3 mL). The organic layer was separated, dried over anhydrous magnesium sulfate, and concentrated in vacuo to give a colorless solid, which was purified by column chromatography (200 mesh silica gel, ethyl acetatehexane = 1:10) to give **5a** (22 mg, 72% yield). Mp: 140 °C (dec). TLC: R_f 0.33 (hexane-ethyl acetate = 10:1). 1H NMR (200 MHz, CDCl₃) δ 1.28 (s, 24H), 5.85 (d, J = 3.9 Hz, 2H),

- 5.96 (d, J = 3.9 Hz, 2H); 13 C NMR (50 MHz, CDCl₃) δ 24.8, 83.5, 130.6; IR (nujol) 1460, 1375, 1340, 1300, 1277, 1218, 1120, 1102, 959, 880, 847, 740, 682 cm $^{-1}$; MS (70 eV) m/z 307 (M $^{+}$ +1, 7.0), 306 (M $^{+}$, 40.0), 305 (M $^{+}$ -1, 19.6), 291 (M $^{+}$ -Me, 8.5), 165 (100). Anal. Calcd for C $_{16}$ H $_{28}$ B $_{2}$ O $_{4}$: C, 62.80; H, 9.22. Found: C, 62.53; H, 9.42.
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