SYNLETT Spotlight 4

This feature focuses on a reagent chosen by a postgraduate, highlighting the uses and preparation of the reagent in current research

Palladium(II) Acetate³

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The X-ray structure of palladium(II) acetate reveals a trimeric molecule with approximate D_3h symmetry in which the Pd atoms are joined by acetate bridges. The reactivity of the molecule varies depending on the nature of cocrystallizing solvent. This reagent is stable, soluble in common organic solvents, and most often used as a catalyst precursor for Pd(0)-catalyzed processes.

Preparation:

1) Palladium(II) acetate can be obtained as brown crystals from the reaction of palladium(II) nitrate with acetic acid.³

2) It can also be prepared from metallic Pd by dissolving it in acetic acid containing nitric acid.³

$$Pd \xrightarrow{AcOH} Pd(OAc)_2$$

$$1$$

Abstracts

A) An efficient procedure for the palladium-catalyzed oxidation of primary and secondary allylic and benzylic alcohols to aldehydes and lactones is achieved using $\rm O_2$ gas as the stoichiometric oxidant.⁴

B) Acetals of acetophenones are directly synthesized by arylation of commercially available 2-ethenyloxyethanol and aryl triflates in the presence of a catalytic amount of **1** and DPPP.⁵

C) The *N*-Arylation of azoles and imines using DPPF-ligated palladium can be accomplished. Imines are markedly more reactive in these transformations.⁶

References

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