



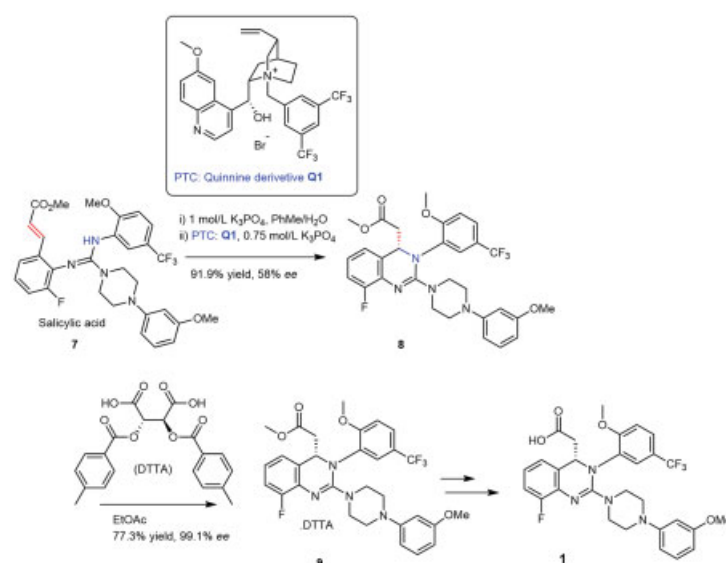
# Enantioselective aza-Michael Cyclization Reaction Catalyzed by Quinine-Derived Monoquaternary Ammonium Salts: an Effective Route to Synthesize Letermovir

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## Abstract

### Keywords

- ▶ letermovir
- ▶ monoquaternary ammonium salts
- ▶ phase-transfer catalyst
- ▶ enantioselective
- ▶ aza-Michael cyclization
- ▶ racemization

A series of mono- or bis-quaternary ammonium salts derived from cinchonidine or quinine was synthesized and screened as potent phase-transfer catalysts for the reaction of aza-Michael cyclization, the key step in the synthesis of letermovir. During the reaction of aza-Michael cyclization, the screened monoquaternary ammonium salt quinine derivative Q1 transferred 7 to 8 with 91.9% yield and 58% ee. The application of Q1 was preferred, due to its enantioselectivity, the possibility of reuse, and the lower cost in large-scale preparation. Furthermore, the racemization condition of letermovir enantiomer was also explored for the possibility to develop the resolution/racemization process. With the optimal catalyst Q1 in hand, the synthesis of letermovir may be more convenient and economical in the future.

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## Introduction

Human cytomegalovirus (HCMV) infections remain a prevalent cause for morbidity and mortality in immunosuppressed solid-organ and bone-marrow transplant patients, and the most frequent cause of opportunistic viral infection following transplantation.<sup>1,2</sup> The incidence of HCMV infection in the general population ranges from 36 to 90%, and the seropositivity increases with age and tends to be highest among lower socioeconomic groups.<sup>3,4</sup> Letemovir is a novel antiviral compound currently used for the prevention of HCMV infections. Before the approval of letemovir, the first line of systemic anti-HCMV drugs includes ganciclovir, foscarnet, and cidofovir, which are derivatives of nucleosides and all target the DNA polymerase, yet, associated with significant side effects such as clinically significant myelosuppression,<sup>5,6</sup> leading to the limited use of them in HCMV therapy.<sup>7</sup> Thus, developing a safe and effective agent for HCMV therapy has been an area of unmet need that is urgently needed in cytomegalovirus (CMV) management. In November 2017, letemovir was approved by the U.S. Food and Drug Administration for CMV prophylaxis in adult allogeneic CMV-seropositive hematopoietic stem cell transplant recipients.<sup>8</sup> As a novel 3,4-dihydro-quinazoline-4-yl-acetic acid derivative, letemovir represents a new class of nonnucleoside CMV inhibitors. Unlike other anti-CMV therapies, letemovir disturbs the late stages of viral replication rather than inhibiting the viral DNA polymerase,<sup>9,10</sup> with no significant myelotoxicity or nephrotoxicity appearing.<sup>11</sup>

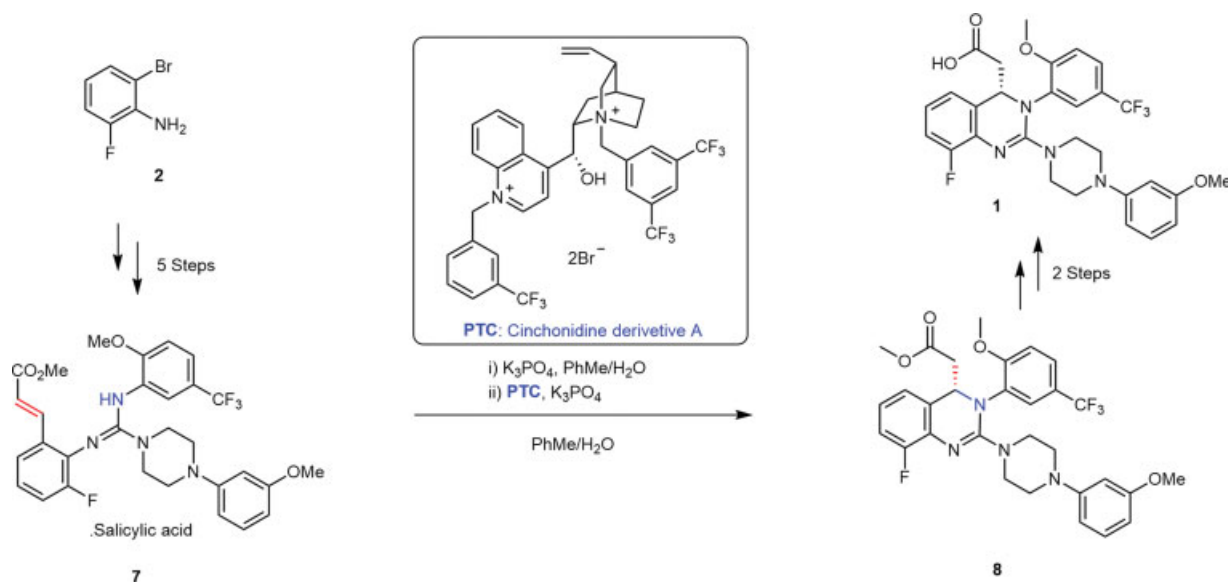
The commercial synthetic route to letemovir (**1**) was reported by Merck and Co., Inc.<sup>12</sup> Letemovir was obtained from **2** through eight steps, including Heck reaction, urea condensation, dehydration, electrophilic substitution, asymmetrical aza-Michael cyclization, chiral resolution, and hydrolysis. The key step is the enantioselective aza-Michael cyclization with bis-quaternary ammonium salt cinchonidine derivative **A** using as the phase-transfer catalyst (PTC)

(**Scheme 1**). PTC **A** is not commercially available and needs to be prepared from a very expensive material, cinchonidine. Moreover, as a di-benzyl-substituted bis-quaternary ammonium salt, PCT **A** is hard to be recycled due to its instability in the reaction and the work-up process. This made us find a better PTC for the replacement.

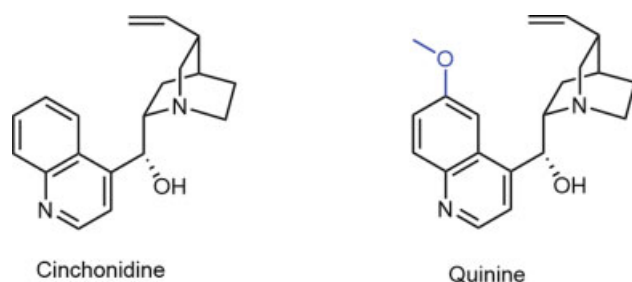
Inspired by our previous work on the synthesis of dezoquine, 17 monoquaternary ammonium salts, derived from cinchonidine or quinine (**Scheme 2**), were screened as the PTCs for the asymmetrical alkylation of 1-methyl-7-methoxy-2-tetralone and some of them were found to function well.<sup>13</sup> As a continuation, it is interesting to evaluate their effect on the asymmetrical aza-Michael cyclization. Herein, we reported an effective and economical route to synthesize letemovir using a quinine-derived monoquaternary ammonium salt (**Q1**) as the PTC in the reaction of aza-Michael cyclization. When we used **Q1** as the catalyst, the cost of the catalyst was far cheaper in comparison to that of PCT **A**; furthermore, the catalyst can be reused and the *ee* value of the target product (**8**) was 58%. Thus we consider that **Q1** was a better catalyst for the aza-Michael cyclization.

## Materials and Methods

All solvents and reagents were purchased from commercial sources and were used without further purification except compound **2**, which was used after distillation. **2** (2-bromo-6-fluorobenzenamine) was purchased from Fuxin Jintelai Fluorin Chemical Co., Ltd. Quinine and cinchonidine were purchased from Shanghai Macklin Biochemical Co., Ltd. (+)-di-*p*-Toluoyl-*D*-tartaric acid (*D*-(+)-DTTA) was purchased from Shanghai Haoyuan Chemexpress Co., Ltd. Melting points were determined on a Büchi Melting Point M-565 apparatus. <sup>1</sup>H spectra were recorded using a Bruker 400 MHz spectrometer with chlorotrimethylsilane as an internal standard. Mass spectra were recorded with a quadrupole time-



Scheme 1 Commercialized process for the synthesis of letemovir (**1**).



Scheme 2 Structures of cinchonidine and quinine.

of-flight mass spectrometer using electrospray positive ionization (ESI+). The enantiomeric ratio was determined by high-performance liquid chromatography (HPLC, Agilent 1260) with a chiral column (CHIRALPAK IA). The eluent was the solution of hexane/isopropyl alcohol (75:25) with 0.1% diethylamine being added, and the detection wavelength was 254 nm. Specific rotations were determined on a Rudolph Research Analytical automatic polarimeter IV. All reactions were monitored by thin layer chromatography, which were performed on silica gel GF254. Column chromatography was performed on silica gel (200–300 mesh) purchased from Qingdao Ocean Chemical Company of China.

**General Procedure for the Preparation of Methyl-(S)-2-(8-fluoro-3-(2-methoxy-5-(trifluoromethyl)phenyl)-2-(4-(3-methoxyphenyl)piperazin-1-yl)-3,4-dihydroquinazolin-4-yl)acetate (2S,3S)-2,3-bis((4-methylbenzoyl)oxy)succinate ethyl acetate solvate (9)**

To a stirred mixture of toluene (700 mL) and **7** (70 g, 96.68 mmol) was added  $K_3PO_4$  aqueous solution (1 mol/L, 145 mL) under an ice bath. After 20 minutes, the mixture became clarified. The organic layer was separated, and  $K_3PO_4$  aqueous solution (0.43 mol/L, 272 mL) was added at 0 to 5°C followed by the addition of **Q1** (3.05 g, 4.83 mmol). After warming to room temperature, the reaction solution was stirred for 5 hours until **7** was consumed, and then glycolic acid solution (1 mol/L, 145 mL) was added. The resulting mixture was stirred for 10 minutes at 45°C, and then cooled down. The organic layer was separated and evaporated to give **8** as yellow oil (52.1 g, yield: 91.9%; HPLC chiral purity: **8a:8b**/78:22).

The oil **8** was dissolved in ethyl acetate (1,000 mL), and then *D*-(+)-DTTA (34.34 g, 88.9 mmol) was added. The mixture was stirred overnight and filtered. The filtrate was washed with ethyl acetate (100 mL) and dried at 75°C to afford **9** (66.83 g, yield: 77.3%). Mp 128–129°C.  $^1H$  NMR (400 MHz,  $DMSO-d_6 + D_2O$ )  $\delta$  7.91–7.89 (d,  $J = 8.0$  Hz, 4H), 7.58–7.55 (dd,  $J_1 = 12.0$  Hz,  $J_2 = 4.0$  Hz, 1H), 7.41–7.39 (d,  $J = 8.0$  Hz, 4H), 7.28–7.26 (d,  $J = 8.0$  Hz, 1H), 7.07 (t,  $J = 12.0$  Hz, 2H), 6.93–6.87 (m, 2H), 6.47–6.44 (dd,  $J_1 = 8.0$  Hz,  $J_2 = 4.0$  Hz, 1H), 6.40–6.36 (m, 2H), 5.82 (s, 2H), 4.94 (t,  $J = 8.0$  Hz, 1H), 4.02 (q,  $J = 12.0$  Hz, 2H), 3.87 (brs, 3H), 3.69 (s, 3H), 3.64 (s, 3H), 3.47–3.37 (m, 5H), 2.96–2.94 (m, 2H), 2.85–2.81 (m, 2H), 2.57 (m, 1H), 2.41 (s, 6H), 1.99 (s, 2H), 1.24 (s, 1H), 1.18

(t,  $J = 8.0$  Hz, 2H). MS (ESI+)  $m/z$  calcd. for  $C_{30}H_{30}F_4N_4O_4$   $[M + H]^+$  587.22, found 587.21.

**General Procedure for the Preparation of (S)-2-(8-Fluoro-3-(2-methoxy-5-(trifluoromethyl)phenyl)-2-(4-(3-methoxyphenyl)piperazin-1-yl)-3,4-dihydroquinazolin-4-yl)acetic acid (Letermovir, 1)**

DTTA salt **9** (40.0 g, 40.12 mmol) was slurried in ethyl acetate (400 mL), and then  $Na_2HPO_4$  aqueous solution (0.5 mol/L, 206 mL) was added. The mixture was stirred for 0.5 hours. The separated organic phase was concentrated to get a white bubble-shaped solid, which was dissolved in MeOH (240 mL), and in NaOH aqueous solution (1 mol/L, 120 mL). The mixture was stirred at room temperature for 1.5 hours, acidified with HCl (3 mol/L, 30 mL) to pH value of 6 to 7, and then extracted with ethyl acetate (200 mL  $\times$  3). The combined organic phase was washed with brine, dried with  $MgSO_4$ , and concentrated to afford letermovir (22.6 g; yield: 96%; HPLC achiral purity: 99.5%, and chiral purity: 99.4%). Mp 117–119°C.  $^1H$  NMR (400 MHz,  $CDCl_3 + D_2O$ )  $\delta$  7.51–7.49 (d,  $J = 8.0$  Hz, 1H), 7.28 (s, 1H), 7.13 (t,  $J = 8.0$  Hz, 1H), 7.03–6.96 (m, 3H), 6.88–6.86 (d,  $J = 8.0$  Hz, 1H), 6.44–6.38 (m, 2H), 6.32 (t,  $J = 4.0$  Hz, 1H), 4.99 (t,  $J = 8.0$  Hz, 1H), 3.76 (s, 6H), 3.56 (s, 4H), 2.96–2.95 (m, 4H), 2.88 (s, 1H), 2.72 (dd,  $J_1 = 16.0$  Hz,  $J_2 = 8.0$  Hz, 1H). MS (ESI+)  $m/z$  calcd. for  $C_{29}H_{28}F_4N_4O_4$   $[M + H]^+$  573.20, found 573.24.

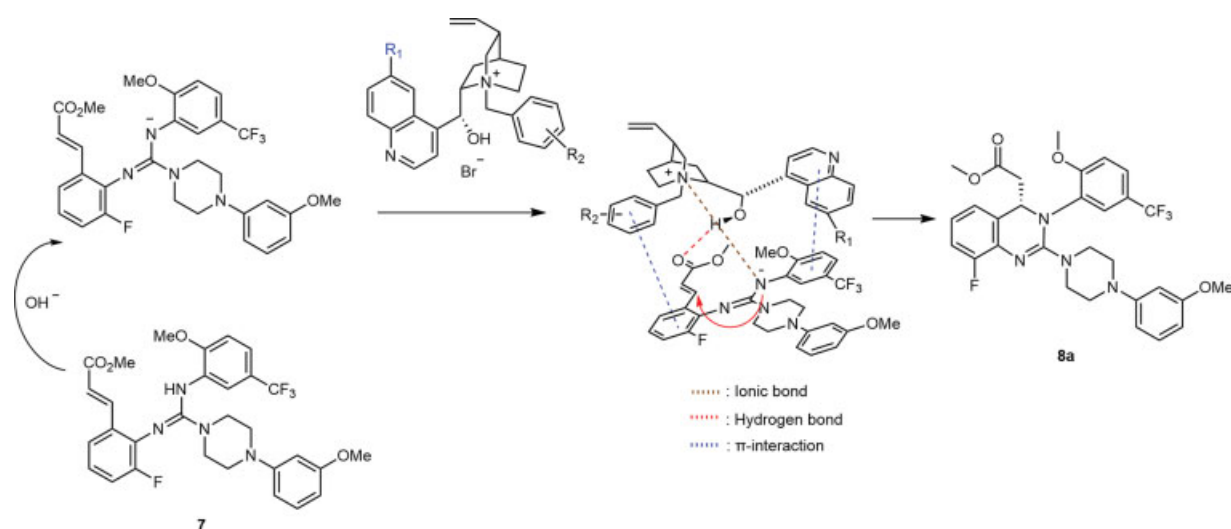
## Results and Discussion

### PTC Screening

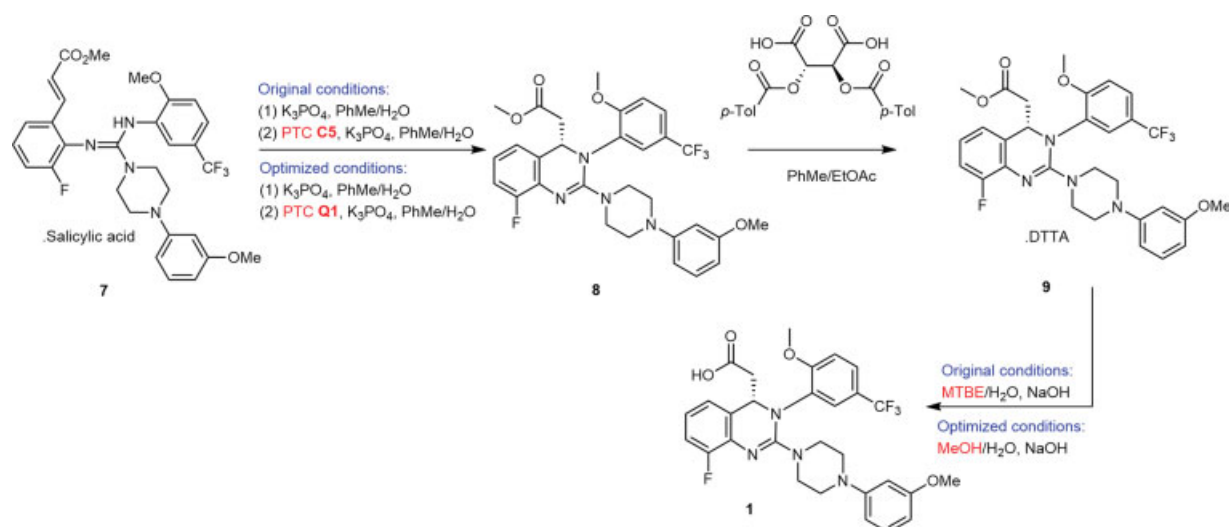
The literature suggested that alkaloids bearing benzyl with electron-withdrawing groups are beneficial for the enantioselective aza-Michael addition.<sup>12</sup> Therefore, eight monobenzyl quaternary ammonium salts from cinchonidine (**C1–C4**) and quinine (**Q1–Q4**) were adopted to evaluate the structure–enantioselectivity relationship, and six bis-quaternary ammonium salts (**C5–C7** and **Q5–Q7**) were used for comparison.

The aza-Michael cyclization of compound **7** was performed in toluene/ $H_2O$  with the base of  $K_3PO_4$  and the catalysis of PTC. When compound **7** was completely consumed, the target product was purified by flash chromatography followed by enantioselectivity analysis through a chiral HPLC determination. The results are listed in **Table 1**.

Our data suggested that monoquaternary ammonium salts, both from cinchonidine (**C1–C4**) and quinine (**Q1–Q4**), had a good effect on enantioselectivity; for example, both entry 1 and entry 8 gave the enantiomeric ratio of **8a:8b** at 79:21. This result was meaningful, because the commercial price of quinine is about one-20th of that of cinchonidine. Besides, the results further demonstrated that the benzyl substituted with two trifluoromethyl was better than that substituted with one trifluoromethyl (**C1** versus **C2** or **C3**; **Q1** versus **Q2** or **Q3**), so compound **Q1** could be considered as the potential PTC. The screening results also suggested that other electron-withdrawing groups in benzyl such as 3,4-dichloro did not improve the enantiomeric ratio.



Scheme 3 Proposed catalytic mechanism of the stereoselective aza-Michael cyclization.



Scheme 4 The optimized process of letemrovir (**1**) in comparison to a reported study.<sup>12</sup>

Evidence suggested that the bisbenzyl quaternary ammonium salt **C5**, a cinchonidine derivative, was reported to have good enantioselectivity, which was further confirmed in entry 5 (8a:8b = 73:27).<sup>12</sup> However, other bis-quaternary ammonium salts **C6**, **C7**, **Q6**, and **Q7** gave the worse enantiomeric ratio of the reaction in comparison to that of **Q5**. Furthermore, we found that the bi-quaternary ammonium salts were hard to be recovered, and the preparation of them was also complex. Bis-quaternary PTCs, if applied in large scale, would cost much more than the monoquaternary ones. Thus, **Q1** was selected for the following large-scale reaction.

### Proposed Catalytic Mechanism

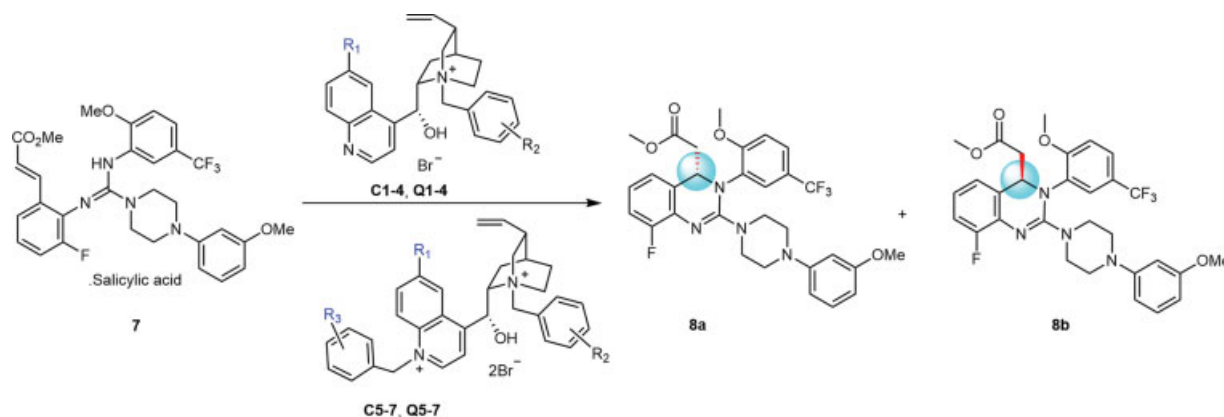
According to the result of the PTC screening, the catalytic mechanism was proposed as shown in **Scheme 3**. Briefly, an anion on the nitrogen of the guanidino moiety of **7** was formed through base treatment, then, the anion intermediate interacted with the catalyst to form the transition state through an ionic bond in the interface of the bi-phase

(toluene/water), as well as a hydrogen bond between the carbonyl oxygen atom of the substrate and the hydroxyl of the catalyst. Meanwhile, the phenyl on the substrate provided the  $\pi$ - $\pi$  interaction with the benzyl of the catalyst. Under the action of the catalyst, N<sup>-</sup> selectively attacked the substrate in the opposite direction to the  $\pi$ - $\pi$  interaction site to complete the stereoselective aza-Michael cyclization.

### Optimized Large-Scale Synthesis of Letemrovir (**1**)

The optimal PTC **Q1** encouraged us to optimize the total synthetic route of letemrovir that was reported by Merck and Co., Inc. **Scheme 4** illustrates an optimized process of letemrovir (**1**) synthesis in comparison to a reported study.<sup>12</sup> Briefly, compound **7** (as a salicylic acid salt) was obtained from 2-bromo-6-fluoroaniline through five steps and underwent the stereoselective aza-Michael cyclization under the catalysis of **Q1**. After DTTA resolution, **9** was hydrolyzed in MeOH/H<sub>2</sub>O to give the target **1**.

**Table 1** Screening of PTCs for the asymmetry aza-Michael cyclization of **7**



Entry	Catalyst	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	Yield <sup>a</sup>	8a:8b <sup>b</sup>
1	C1	H	3,5-CF <sub>3</sub>	–	80.0%	79:21
2	C2	H	3-CF <sub>3</sub>	–	92.0%	71:29
3	C3	H	4-CF <sub>3</sub>	–	76.0%	65:35
4	C4	H	3,4-Cl	–	68.3%	71:29
5	C5	H	3,5-CF <sub>3</sub>	3-CF <sub>3</sub>	77.0%	73:27
6	C6	H	3,5-CF <sub>3</sub>	3,5-CF <sub>3</sub>	81.5%	64:36
7	C7	H	3-CF <sub>3</sub>	3-CF <sub>3</sub>	66.7%	59:41
8	Q1	–OCH <sub>3</sub>	3,5-CF <sub>3</sub>	–	91.9%	79:21
9	Q2	–OCH <sub>3</sub>	3-CF <sub>3</sub>	–	84.9%	69:31
10	Q3	–OCH <sub>3</sub>	4-CF <sub>3</sub>	–	91.1%	65:35
11	Q4	–OCH <sub>3</sub>	3,4-Cl	–	96.2%	69:31
12	Q5	–OCH <sub>3</sub>	3,5-CF <sub>3</sub>	3-CF <sub>3</sub>	67.5%	71:29
13	Q6	–OCH <sub>3</sub>	3,5-CF <sub>3</sub>	3,5-CF <sub>3</sub>	63.3%	63:37
14	Q7	–OCH <sub>3</sub>	3-CF <sub>3</sub>	3-CF <sub>3</sub>	74.9%	62:28

Abbreviation: PTCs, phase-transfer catalysts.

<sup>a</sup>Isolated as a mixture of **8a** and **8b**.

<sup>b</sup>Enantioselectivity was determined by chiral HPLC analysis.

**Table 2** Racemization of the enantiomer

Entry	Base	Solvent	Temperature (°C)	Time (h)	Specific rotation	Result
1	Sodium methoxide <sup>a</sup> , 2 equiv.	Acetonitrile	Reflux	72	221.2 (c = 1, MeOH) <sup>b</sup>	No racemization
2	Sodium methoxide <sup>a</sup> , 2 equiv.	Methanol	Reflux	48	235.2 (c = 1, MeOH) <sup>b</sup>	No racemization
3	DBU, 2 equiv.	Acetonitrile	Reflux	72	237.3 (c = 1, MeOH) <sup>b</sup>	No racemization
4	DBU, 3 equiv.	DMSO	65	16	268 (c = 1, MeOH) <sup>b</sup>	No racemization
5	KOH, 2 equiv.	DMSO	100	10	6.5 <sup>c</sup>	Racemization
6	KOH, 4 equiv.	DMSO	100	10	–	Several impurities

<sup>a</sup>30% solution in methanol.

<sup>b</sup>c = 1 means 1 g/100mL.

<sup>c</sup>R:S = 52.7: 47.2, determined by chiral HPLC analysis.

However, some trouble was encountered in practice. During dissociation of the freebase from **7** in the presence of base K<sub>3</sub>PO<sub>4</sub> at ca. 30°C, some percentage (10–30%) of the background aza-Michael cyclization already taken place before the

addition of PTC. It was pleased to find that if this dissociation was performed at 0 to 5°C, the background reaction could be eliminated. In the presence of **Q1** and K<sub>3</sub>PO<sub>4</sub>, **8** was obtained from **7** in a yield of 68.3% with 58% ee. After the salt formation

with *D*-(+)-DTTA in ethyl acetate, the *ee* value increased to 99.3%. The catalyst was recovered by column chromatography from the mother liquor and could be reused in the same reaction without the loss of yield or *ee* value.

In the hydrolysis step, a long reaction time (at least 20 hours) was required when the hydrolysis was performed in NaOH aqueous solution at 60°C, according to a reported study,<sup>12</sup> which increased the risk of the racemization of the product. To our delight, when the reaction solution was replaced with NaOH aqueous solution/MeOH (*v/v* = 1:2), the reaction could be performed at room temperature and the reaction time could be shortened to approximately 1 hour, giving **1** in a yield of 95%, an achiral purity of 99.5%, and a chiral purity of 99.4%.

A small amount of racemization was also observed in the hydrolysis step, especially if the reaction lasted for a long time. To verify the possibility of the racemization of the product and develop the process to reuse the enantiomer of letermovir, experiments on the racemization of the enantiomer were further designed. The mother liquor (**8a:8b**/9.3:90.7) from *D*-(+)-DTTA salt formation was recovered and hydrolyzed into the acid, which was used for the racemization, and the result is included in ▶Table 2. Our data suggested that sodium methoxide in acetonitrile or methanol under reflux did not realize the racemization. 1,8-Diazabicyclo[5.4.0]undec-7-ene (DBU) was not the ideal base. Finally, racemization could be performed under the condition of 2 equiv. of KOH in DMSO at 100°C (entry 5) with *R:S* = 52.7: 47.2 identified by chiral HPLC. This result laid the foundation to develop a recycling process of resolution/racemization to reuse the enantiomer of letermovir.

## Conclusion

In summary, an optimal catalyst **Q1** (quinine-derived monoquaternary ammonium salt) was explored, due to its good enantioselectivity, easy preparation from cheap material, as well as the possibility of reuse. With the use of the selected **Q1**, the synthesis of letermovir would be more convenient and economical. Furthermore, the hydrolysis (the last step) of the total synthesis route of letermovir was also improved to ensure the purity of the target product. The racemization condition of the enantiomer of letermovir was explored for the possibility to develop the resolution/ racemization process so that the enantiomer could be completely used.

## Supplementary Information

Detailed information for the synthesis of the catalysts **C1–C8**, **Q1–Q7**, as well as the chiral HPLC diagrams of **8** is provided and available online only.

## Conflict of Interest

None.

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