A Second-Generation Chemoenzymatic Total Synthesis of Platencin

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Abstract A total synthesis of the potent antibacterial agent platencin is described. The reaction sequence used involves, as starting material, an enantioselectively pure cis-1,2-dihydrocatechol derived from the whole-cell biotransformation of iodobenzene. Simple chemical manipulations of this metabolite provide a triene that engages in a thermally promoted intramolecular Diels–Alder reaction to establish the octahydro-2H-1,4a-ethanonaphthalene core of platencin. As part of a range of programs directed towards the identification of clinically useful analogues, various studies have focussed on establishing total syntheses of platensimycin (1) and platencin (2). In 2008, we reported a synthesis of the tricyclic enone 3 (Figure 2), an advanced intermediate in Nicolaou’s original total synthesis6a of platencin (2). The starting material used in our study was the enantioselectively pure cis-1,2-dihydrocatechol (4) obtained through a whole-cell biotransformation of iodobenzene using a genetically engineered microorganism E. coli JM109 (pDTG601) that overexpresses the enzyme toluene dioxygenase.11 Over a number of simple steps, including a Negishi cross-coupling reaction, compound 4 was converted into a triene that participated in a thermally induced intra-
molecular Diels–Alder (IMDA) reaction, thus affording the tricarbocyclic framework embodied within compounds 2 and 3.

The retrosynthetic analysis employed in the present study is shown in Scheme 1 and anticipated that target 2 could be prepared, through straightforward functional group interconversions (FGI), from ester 5 that would itself be formed by employing, as a first step, our previously reported IMDA reaction of the triene 6 and then elaborating the resulting adduct into the pivotal subtarget 5. In our original studies we showed that the substrate 6 required for the crucial IMDA reaction could be obtained through Negishi cross-coupling of the iodide 7 with the easily generated organozinc species 8. Compound 7 is the readily accessible12–14 acetonide derivative of cis-1,2-dihydrocatechol (4),15 while the iodide precursor to compound 8 can be produced in just four steps from diallyl alcohol.16

A difficulty associated with this approach was the limited amount of intermediate 3 available by such means. Furthermore, introducing the required C4 methyl and propionic acid groups in an efficient and stereocontrolled manner proved problematic. In an attempt to address these issues, we recently reported an alternate IMDA-based approach that delivered a more highly functionalised form of compound 3 and one that could be converted, over a further thirteen steps, into platencin. However, significant functional group incompatibilities were encountered in this first chemoenzymatic total synthesis of the title natural product. Given the deficiencies associated with both of our previous routes we pursued a second-generation chemoenzymatic total synthesis. As a result, we have established, and now report, a more concise as well as a regio- and stereoregulated pathway to compound 2 that exploits key elements of our earlier approach to enone 3.

The opening stage of the present synthesis involved, as we have shown previously but repeat here for the sake of completeness, conversion (Scheme 2) of the known8 and racemic iodoide 9 into the organozinc species 8 by successive treatment of the former compound with t-butyllithium and then zinc iodide between −78 and 18 °C followed by cross-coupling of the latter with iodide 7 in the presence of Pd(Ph₃P)₄. By such means, the triene 67 was obtained as a 1:1 mixture of diastereoisomers in 73% yield (from 9). While rather acid sensitive, when a toluene solution of compound 6 containing butylated hydroxytoluene (BHT – added to suppress oxidative degradation of the substrate) was heated at 110 °C for 16 hours, the corresponding mixture of diastereoisomeric Diels–Alder adducts 107–16 was obtained in 89% combined yield.

The means for converting the diastereoisomeric forms of compound 10 into the ester 5 are shown in Scheme 3 and involved, as a first step, treating the former compound with tetra-n-butylammonium fluoride (TBAF) in THF at 18 °C. The α-epimeric form of compound 107 (embodying an endo-configured TBSO group) was converted into the corresponding alcohol 117 over 56 hours and in 86% yield, while the notionally more hindered α-epimer7 took just 16 hours to react under the same conditions and afforded the anticipated alcohol in 97% yield.

Hydrogenation of alcohol 11 (as a mixture of epimers) at atmospheric pressures using 10% palladium on carbon as catalyst and ethanol as solvent afforded the corresponding mixture of saturated alcohols 1217 in (97% combined yield) and this was oxidised to the corresponding ketone 1317 (95%) using pyridinium dichromate (PDC). In order to generate the enone required for the twofold α-alkylation reaction that would deliver subtarget 5, compound 13 was treated with trimethylsilyl triflate (TMSOTf) in the presence of trimethylamine. The ensuing mixture of silyl enol ethers was exposed to o-iodoxybenzoic acid (IBX) in the presence of 4-methoxypyridine-N-oxide (MPO)18 and thereby affording a (flash) chromatographically separable mixture of enones 14 (21%) and 15 (65%). The spectroscopic data derived from these products were in complete accord with the assigned structures, and that of the former was confirmed.
R. N. Muhammad et al.

by single-crystal X-ray analysis.\textsuperscript{19} Treatment of compound 15 with potassium hexamethydisilazide (KHMDS) and trapping of the ensuing enolate at \(-78^\circ\text{C}\) with methyl iodide afforded compound 16 (78%), and the structure of this product was also confirmed by single-crystal X-ray analysis.\textsuperscript{19} Following protocols reported by Nicolaou et al.,\textsuperscript{6b} compound 16 was treated with \(\text{t}-\text{BuOK}\) and \(\text{t}-\text{butyl acrylate}\), producing a mixture of the epimeric esters 17 (17%) and 5 (68%) that could only be separated by (semi-preparative) HPLC techniques.

The reaction sequence used to convert ester 5 into platencin (2) is shown in Scheme 4 and parallels that employed in our first-generation route to the natural product.\textsuperscript{5} Thus, the acetonide residue associated with compound 5 was cleaved using acid-activated Dowex-50 resin in a metha-

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**Scheme 2** Assembly of the tricyclic core of platencin via the IMDA reaction of triene 6

**Scheme 3** Conversion of the IMDA adduct 10 into ester 5
nol–water–THF mixture at 65 °C and, after 36 hours, a 63% yield (at 83% conversion) of the required diol 18 was realised. Regioselective oxidation of the hydroxyl group remote from the cyclohexenone substructure within compound 18 could be achieved using the sterically demanding oxammonium salt derived from the p-toluenesulfonic acid (PTSA)-promoted disproportionation of 4-acetamido-TEMPO\(^{20}\) and the acyloin 19 (91%) thus obtained. As the first step in the removal of the remaining hydroxyl group within compound 19 this was converted into the corresponding acetate 20 (83%) under standard conditions. Reductive cleavage of the newly introduced ester residue was most conveniently achieved using the readily prepared but distinctly underutilised vanadium(II) complex \([V_2Cl_3(THF)_6][Zn_2Cl_6]\) first introduced for this purpose by Torii and co-workers over two decades ago.\(^{21–23}\) The reaction proceeded rapidly under ultrasonication conditions and afforded dione 21\(^{24}\) in 81% yield. Selective methylation of the nonconjugated ketone carbonyl moiety within compound 21 was achieved using the Wittig reagent under carefully controlled conditions,\(^9\) to afford the previously reported ester 22\(^6\) in 85% yield. In contrast to observations of Nicolaou and co-workers,\(^6\) it was found that the t-butyl ester moiety within this last compound could be cleaved using trifluoroacetic acid (TFA, at 18 °C for 0.5 h) without any accompanying conversion of the exocyclic olefin to its endocyclic isomer. Platencinic acid (23)\(^{6,8,9,24}\) was thus formed in ca. 85% yield. The spectroscopic data derived from this material were in complete accord with those reported by both our group\(^9\) and by others.\(^6,8\) Because we have previously\(^6\) coupled this last compound with aniline 24\(^9\) to form platencin (2) in 42% yield, the present work constitutes a formal synthesis of the title natural product.

The route to platencin (2) described here exploits the capacity, as reported in our earlier studies,\(^7\) of the readily accessible and enantiomerically pure triene 6 to engage in an IMDA reaction and thereby generate the tricarbocyclic framework of the natural product. The value of the work

![Scheme 4](attachment:image.png)

Scheme 4  End game: conversion of ester 5 into platencin (2)
detailed above lies in the defining of a method by which cy-cloadduct 10 can be elaborated to platencinic acid (23), an immediate precursor to platencin and itself a naturally occurring compound. The present route is slightly shorter than our first-generation chemoenzymatic synthesis of platencin. However, it remains wanting because of an inability to install the enone double bond in a completely regiocon-trolled manner and a failure to fully control the stereoselec-tivity of the Michael addition reaction (16 → 5) that establishes the C4 quaternary carbon centre. The second issue is the more problematic because of the need to resort to te-rious HPLC techniques to separate the diastereoiseric product esters 5 and 17 (see Scheme 3). Efforts focussed on addressing such matters as well as exploiting the protocols reported here for the purposes of preparing platencin ana-logues are under investigation.

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Supporting Information

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References and Notes


then diluted with EtOAc (5.0 mL) and passed through a short plug of TLC-grade silica gel that was washed with EtOAc (3 × 5 mL). The combined filtrates were washed with H$_2$O (1 × 10 mL) before being dried (MgSO$_4$), filtered, and concentrated under reduced pressure. The resulting light-yellow oil was subjected to flash chromatography (silica, 1:4 v/v EtOAc–hexane elution) to afford, after concentration of the relevant fractions ($R_f = 0.4$), compound 21$^{1a}$ (75 mg, 81%) as a white foam. A full spectroscopic data set for this compound is provided in the Supporting Information.