STRIP

Max-Planck-Institut für Kohlenforschung, Kaiser-Wilhelm-Platz 1, 45470 Mülheim an der Ruhr, Germany list@kofo.mpg.de

Received: 10.01.2015 Accepted after revision: 31.01.2015 Published online: 24.02.2015

DOI: 10.1055/s-0034-1380409; Art ID: st-2015-d0018-l

Abstract We report a catalytic enantioselective desymmetrization of *meso*-1,2-diols by monosilylation using a chiral enantiopure Brønsted acid as catalyst and hexamethyldisilazane (HMDS) as silyl source.

Key words enantioselective Brønsted acid catalysis, silyl transfer, *meso*-diols, desymmetrization, asymmetric counteranion-directed catalysis (ACDC)

The silylation of alcohols is a widespread transformation in organic synthesis, typically used to introduce protecting groups. Frequently employed protocols involve basic conditions and a silyl chloride or triflate reagent. Recently, the first examples of catalytic asymmetric alcohol silylations have been described. Interestingly, these methods are based on electrophilic silicon reagents employed under basic conditions. We were intrigued to design catalytic cycles that are based on an acidic activation of basic silicon sources, which are susceptible to proto desilylation. Here we report our studies on the Brønsted acid catalyzed desymmetrization of *meso*-diols using hexamethyldisilazane (HMDS) as the silicon source and the chiral spirocyclic phosphoric acid STRIP, recently introduced from this laboratory, as the catalyst.

The enantioselective desymmetrization of *meso*-diols is one of the most prominent strategies to access the corresponding monoprotected derivatives in their enantiopure form. This reaction is commonly realized by stereoselective acyl transfer reactions.³ However, the introduced acyl group may not always be suited as a protecting group in a given synthetic context, thus rendering further functional group transformations necessary.^{2a} As an alternative, highly efficient catalytic enantioselective monosilylations of *meso*-diols have been developed by Hoveyda, Snapper and co-

workers.^{2a-c} In analogy to the most common protocols for the silyl protection of alcohols, which are typically carried out under basic conditions,¹ these reactions involve the use of an enantiopure basic catalyst in combination with an achiral stoichiometric base such as Hünig's base. Despite the predominance of basic conditions for the silyl protection of alcohols, non-enantioselective protocols using acid catalysts in combination with basic silyl transfer reagents have also been developed.^{1,4}

Based on the mechanism of this type of transformation, we hypothesized that it should be possible to render the overall process enantioselective by utilizing a chiral enantiopure Brønsted acid catalyst (Scheme 1).⁵ Accordingly, protonation of the basic silicon source should generate an ion pair consisting of a cationic silylium source accompanied by the enantiopure counteranion.

Scheme 1 Design of a Brønsted acid catalyzed desymmetrization of *meso*-diols

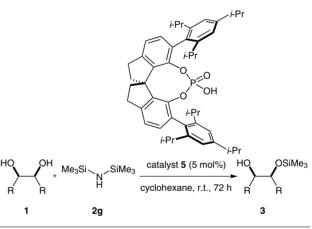
The reaction of this ion pair intermediate with the diol substrate could then potentially proceed enantioselectively via a desymmetrization reaction.

'	a Z			Ja
Entry	Nu-SiR ₃	Equiv	Conv. (%) ^a	er ^b
1	TMSCI 2a	1.2	0	-
2	TMS 2b	1.2	0	-
3	OTMS OMe	1.2	20	63:37
4	TMSCN 2d	1.2	11	63:37
5	OTMS	0.6	42	38.5:61.5
6	N TMS	1.2	8	50:50
7	TMS TMS H	0.6	78	81.5:18.5
8	Et Si Et Si Et	0.6	60	78:22
9	<i>t</i> -Bu <i>t</i>	0.6	0	-
10	Ph Ph Ph Si N H	0.6	21	81:19
11 ^c	TMS TMS H	0.6	83	94:6

^a Determined by ¹H NMR using 1,1,2,2-tetrachloroethane as an internal

We began by studying several prospective silyl transfer reagents utilizing TRIP (4a)⁶ as reference catalyst (Table 1). Comparably non-basic silvlating agents trimethylsilyl chloride (2a) and allyltrimethylsilane (2b) proved to be inefficient for product formation under the reaction conditions tested (entries 1 and 2 in Table 1). However, switching to more basic silyl transfer reagents enabled reactivity with moderate to good conversion, albeit with limited stereoinduction (entries 3-6 in Table 1). Further screening revealed HMDS (2g) and its derivatives as the most promising reagents (entries 7-10 in Table 1). We thus continued our studies with the commercially available and inexpensive HMDS as silyl source and further optimized the reaction conditions.⁷ Through these studies we identified catalyst **5** (STRIP) as best catalyst for this transformation. 6c giving good conversion and stereoinduction in our model reaction (83% conversion and 94:6 er, entry 11 in Table 1).

Table 2 Substrate Scope of the STRIP-Catalyzed Desymmetrization of meso-Diols



Entry	R	Product	Time (d)	Yield (%)	era
1	Ph	3a	3	84	95:5
2 ^b	4-MeC_6H_4	3b	2	96	95:5
3	$3-MeC_6H_4$	3с	7	92	93.5:6.5
4	2-MeC_6H_4	3d	7	91	90:10
5 ^b	4 - i - PrC_6H_4	3e	1	99	92.5:7.5
6 ^{c-e}	$4-FC_6H_4$	3f	10	77	95: 5
7	4-CIC ₆ H ₄	3g	1	87	95.5:4.5
8e	2-CIC ₆ H ₄	3h	5	94	89:11
9	4 -BrC $_6$ H $_4$	3i	6	76	90.5:9.5
10	-(CH ₂) ₄ -	3j	7	99 ^f	75.5:24.5 ^g

^a Determined by HPLC on a chiral stationary phase.

^b Determined by HPLC on a chiral stationary phase.

^c (S)-STRIP (5; see Table 2) was used as catalyst.

 $^{^{\}rm b}$ Reaction was conducted at 40 $^{\circ}\text{C}$

^c Carbon tetrachloride was used as solvent instead of cyclohexane.

d Amount of catalyst 5 used was 10 mol%.

e Compound 2h was used as a silylating reagent instead of 2g.

f Conversion determined by ¹H NMR using 1,1,2,2-tetrachloroethane as an internal standard.

⁹ Determined by GC on a chiral stationary phase.

In summary, we have developed a catalytic asymmetric desymmetrization of *meso*-1,2-diols by monosilylation. Our method employs STRIP (**5**) as enantioselective catalyst, together with HMDS (**2g**) as silyl source. Moderate to good results were obtained with a number of aryl-substituted *meso*-diols. Further studies on the use of this type of Brønsted acid catalyzed enantioselective silylation are currently ongoing in our laboratories.

Acknowledgment

Generous support from the Max-Planck-Society, the European Research Council (Advanced grant 'High Performance Lewis Acid Organocatalysis, HIPOCAT'), the Naito Foundation (fellowship to K.H.), and the Alexander von Humboldt Foundation (fellowship to S.G.) is gratefully acknowledged.

Supporting Information

Supporting information for this article is available online at http://dx.doi.org/10.1055/s-0034-1380409.

References and Notes

- Greene, T. W.; Wuts, P. G. M. Protection for the Hydroxyl Group, Including 1,2- and 1,3-Diols, In Protective Groups in Organic Synthesis; John Wiley & Sons: New York, 1999, 3rd ed., 17–245.
- (2) (a) Zhao, Y.; Rodrigo, J.; Hoveyda, A. H.; Snapper, M. L. Nature (London) 2006, 443, 67. (b) Zhao, Y.; Mitra, A. W.; Hoveyda, A. H.; Snapper, M. L. Angew. Chem. 2007, 119, 8623. (c) Manville, N.; Alite, H.; Haeffner, F.; Hoveyda, A. H.; Snapper, M. L. Nature Chem. 2013, 5, 768. (d) Sheppard, C. I.; Taylor, J. L.; Wiskur, S. L. Org. Lett. 2011, 13, 3794. (e) Weickgenannt, A.; Mewald, M.;

- Oestreich, M. *Org. Biomol. Chem.* **2010**, *8*, 1497. For selected examples of non-catalytic asymmetric silylations, see: (f) Isobe, T.; Fukuda, K.; Araki, Y.; Ishikawa, T. *Chem. Commun.* **2001**, 243. (g) Rendler, S.; Auer, G.; Oestreich, M. *Angew. Chem. Int. Ed.* **2005**, *44*, 7620.
- (3) (a) García-Urdiales, E.; Alfonso, I.; Gotor, V. Chem. Rev. 2004, 105, 313. (b) Enriquez-Garcia, A.; Kundig, E. P. Chem. Soc. Rev. 2012, 41, 7803. (c) Müller, C. E.; Schreiner, P. R. Angew. Chem. Int. Ed. 2011, 50, 6012.
- (4) (a) Morita, T.; Okamoto, Y.; Sakurai, H. Tetrahedron Lett. 1980, 21, 835. (b) Olah, G. A.; Husain, A.; Gupta, B. G. B.; Salem, G. F.; Narang, S. C. J. Org. Chem. 1981, 46, 5212. (c) Veysoglu, T.; Mitscher, L. A. Tetrahedron Lett. 1981, 22, 1299. (d) Bruynes, C. A.; Jurriens, T. K. J. Org. Chem. 1982, 47, 3966. (e) Suzuki, T.; Watahiki, T.; Oriyama, T. Tetrahedron Lett. 2000, 41, 8903. (f) Firouzabadi, H.; Iranpoor, N.; Amani, K.; Nowrouzi, F. J. Chem. Soc., Perkin Trans. 1 2002, 2601. (g) Shaterian, H. R.; Shahrekipoor, F.; Ghashang, M. J. Mol. Catal. A.: Chem. 2007, 272, 142
- (5) (a) Mahlau, M.; List, B. Isr. J. Chem. 2012, 52, 630. (b) Phipps, R. J.; Hamilton, G. L.; Toste, F. D. Nature Chem. 2012, 4, 603. (c) Mahlau, M.; List, B. Angew. Chem. Int. Ed. 2013, 52, 518. (d) Akiyama, T. Chem. Rev. 2007, 107, 5744. (e) Kampen, D.; Reisinger, C. M.; List, B. Chiral Brønsted Acids for Asymmetric Organocatalysis, In Asymmetric Organocatalysis; Vol. 291; List, B., Ed.; Springer: Berlin/Heidelberg, 2009, 1–37.
- (6) (a) Adair, G.; Mukherjee, S.; List, B. Aldrichimica Acta 2008, 41, 31. (b) Klussmann, M.; Ratjen, L.; Hoffmann, S.; Wakchaure, V.; Goddard, R.; List, B. Synlett 2010, 2189. (c) For the first report of STRIP, see: Čorić, I.; Müller, S.; List, B. J. Am. Chem. Soc. 2010, 132, 17370.
- (7) For further screening of reaction conditions and catalysts, see the Supporting Information.
- (8) The following procedure is representative: To a solution of catalyst 5 (3.6 mg, 5.00 µmol) in cyclohexane (1.0 mL) HMDS (2g, 12.5 µL, 0.0600 mmol) was added at r.t. and the resulting mixture was stirred for 2 h, after which, meso-diol 1a (21.4 mg, 0.100 mmol) was added. The reaction was judged to be complete when the reaction mixture turned into a clear, homogeneous solution. The solvent was removed by evaporation and the crude mixture was purified by silica gel column chromatography (hexane-EtOAc, 95:5), giving 3a (24.0 mg, 84%) as a colorless solid. ¹H NMR (500 MHz, CD_2Cl_2): $\delta = 7.22-7.30$ (m, 10 H), 4.74 (d, J = 5.0 Hz, 1 H), 4.71 (d, J = 5.0 Hz, 1 H), 2.39 (br, 1 H),-0.09 (s, 9 H). ¹³C NMR (125 MHz, CD_2Cl_2): δ = 141.6, 141.5, 128.3, 128.2, 128.1, 128.0, 127.9, 127.7, 79.6, 78.9, 0.00. HRMS (ESI, +ve): m/z [M + Na] calcd for $C_{17}H_{22}NaO_2Si$: 309.1287; found: 309.1281. HPLC (Chiralcel OJ-H, heptane-i-PrOH = 95:5, flow rate = 0.5 mL/min, λ = 206 nm): $t_{R(major)}$ = 12.6 min, $t_{R(minor)}$ = 16.1 min.