Category

Metal-Catalyzed **Asymmetric** Synthesis and Stereoselective Reactions

Key words

iridium

hydrogenation

alkenyl boronates

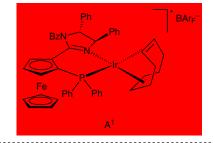
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A Practical Synthetic Approach to Chiral (α-Chloroalkyl)boronic Esters via Iridium-Catalyzed Chemoselective Hydrogenation of Chloro-Substituted Alkenyl Boronates Synthesis 2013, 45, 2824-2831.

Enantioselective Hydrogenation of Chloro-Substituted Alkenyl Boronates

Overall transformation:

$$\begin{array}{c} R \\ CI \\ \end{array} \begin{array}{c} A^1 \text{ (S/C} = 7-25) \\ H_2 \text{ (5-30 bar)} \\ \hline CH_2 \text{CI}_2 \text{ (2.5-3 mL), 50 °C} \\ \hline \textbf{0.18-0.22 mmol scale} \end{array} \begin{array}{c} R \\ BPin \\ \end{array} \begin{array}{c} R \\ BPin \end{array}$$



70% yield A, 88% ee 91% yield A, 89% ee

Significance: Enantioenriched (α-chloroalkyl)boronic esters represent an important class of building blocks frequently used in organic synthesis. These chiral halogenated compounds are most commonly synthesized via Matteson-type rearrangements which use stoichiometric lithiated reagents, leading to stoichiometric waste streams. Transition-metal-catalyzed asymmetric hydrogenation has become the premier method for setting diversely substituted stereocenters with high enantiomeric excess under more environmentally friendly conditions. The authors report the synthesis of enantioenriched (α-chloroalkyl)boronic esters via the iridium-catalyzed asymmetric hydrogenation of chloro-substituted alkenyl boronates.

SYNFACTS Contributors: Mark Lautens, David A. Petrone Synfacts 2014, 10(1), 0060 Published online: 13.12.2013 **DOI:** 10.1055/s-0033-1340445; **Reg-No.:** L15513SF

Comment: The authors present a protocol for the enantioselective iridium-catalyzed hydrogenation of chloro-substituted alkenyl boronates using a ferrocenyl phosphinoimidazoline ligand. Good to excellent yields and enantioselectivities were obtained for both alkyl- and aryl-bearing substrates. Substrate-to-catalyst ratios (S/C) of up to 25 and hydrogen pressures as low as 5 bar are reported. This catalytic system benefits from a low amount of dechlorinated products resulting from overreduction.

65% yield A, 94% ee

73% yield A, 88% ee