Insert, Cyclize, Eliminate: Polyimides from Imidates and Isocyanates

### Representative transformation:

**1**

![ Representative transformation: ](image)

**2**

99% yield

**3**

![ Representative transformation: ](image)

**4**

52% yield

**5a**

R = C₆H₄Hex

60% yield

**5b**

R = C₆H₄(CH₂)₁₁Me

68% yield

**6a**

Mₙ = 4400, Mₘ/Mₙ = 2.87, T₅₀/N₂ = 362 °C (5% wt loss temperature, 10 °C/min)

**6b**

Mₙ = 6200, Mₘ/Mₙ = 2.97, T₅₀/N₂ = 328 °C (5% wt loss temperature, 10 °C/min)

### Significance:

Polyimides find application in aerospace engineering and electronic materials because of their inherent rigidity, strength, heat-resistance, and electrical non-conductance. However, their low solubility in organic solvents limits processability. The authors hypothesized that the solubility could be improved by switching the carbonyl group into an imino group functionalized with pendant alkyl chains. They confirmed their hypothesis by developing a mild, atom-economic rhenium-catalyzed transformation that involves sequential C–H activation, intramolecular nucleophilic cyclization, and elimination of methanol.

### Comment:

The key step in the reported methodology is the restoration of the C=N double bond by elimination of a small molecule (methanol). The efficacy of this tactic is evidenced by the good yields obtained throughout a diverse substrate scope and on a gram-scale reaction (80% yield). This is the first example of polyimide derivatives having imino groups; the polymers obtained are highly soluble in toluene, tetrahydrofuran, dichloromethane, and chloroform. It will be important to explore the mechanical properties of this new class of polyimides to determine how they compare to their less soluble, carbonyl-containing counterparts.