Application of 3-Bromo-3-ethylazetidines and 3-Ethylideneazetidines for the Synthesis of Functionalized Azetidines

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Abstract: The synthetic utility of 3-bromo-3-ethylazetidines has been demonstrated by the straightforward preparation of 3-alkoxy-, 3-aryloxy-, 3-haloalkyl-3-azetidines and dehydrohalogenation of systems so far. The two main literature approaches to in-
For example, the reactions of 3a,b with oxygen nucleophiles, such as methoxide [NaBH₄ (2 equiv), MeOH, Δ, 2 d], phenoxide [PhOH (2.2 equiv), K₂CO₃ (5 equiv), MeCN, Δ, 1 d], sodium acetate [NaOAc (5 equiv), MeCN, Δ, 3–5 d] and potassium hydroxide [KOH (5 equiv), H₂O/CH₂Cl₂ (7:1), Δ, 3–5 d] and potassium hydroxide [KOH (5 equiv), H₂O/CH₂Cl₂ (7:1), Δ, 15–20 h], provided the corresponding 3-alkoxy-, 3-aryloxy-, 3-acetoxy- and 3-hydroxy-3-ethylazetidines 6a–d (Nu = OMe, OPh, OAc, OH), respectively (Scheme 2). The reaction of 3a with n-propylamine furnished 3-propylaminoazetidine 6e [Nu = n-PrNH; n-PrNH₂ (5 equiv), MeCN, Δ, 1 d], while azetidines 3a,b gave 3-cyanoazetidines 6f,g on treatment with potassium cyanide [KCN (1.5 equiv), MeCN, Δ, 1 d].

Further reaction of azetidine-3-carbonitrile 6f with KOH (5 equiv) in EtOH/H₂O (10:1) under microwave irradiation (10 min, 150 °C, 150 W) resulted in the major compound (77%), accompanied by a small amount of the corresponding amino acid 8 (19%; Scheme 2). The formation of carboxylic acid 8 was evidenced upon neutralization of the reaction mixture to pH 7 using 1 M HCl. Attempts to develop an effective synthesis of amino acid 8 by prolonging the reaction time (up to 2 h) proved to be unsuccessful.

In summary, the above-described findings acknowledge the suitability of 3-bromo-3-ethylazetidines as substrates for nucleophilic substitutions by different oxygen-, nitrogen- and carbon-centered nucleophiles.

In the next part of our studies, the eligibility of 3-bromo-3-ethylazetidines as substrates for the preparation of the corresponding new 3-ethylideneazetidines was investigated. Whereas the dehydrobromination of 3-bromo-1-tert-butylazetidine 5 utilizing t-BuOK (1.5 equiv) in THF afforded 1-tert-butyl-3-ethylideneazetidine (9) in a good yield (Scheme 3), the synthesis of 1-arylmethyl-3-ethylideneazetidines 10a,b starting from 3-bromo-3-ethylazetidines 3a,b was not as straightforward as initially anticipated, and several attempts were performed to optimize the reaction conditions. Treatment of azetidine 3a with different bases such as t-BuOK, LDA and NaH in THF at room temperature or under reflux gave no reaction, and addition of t-BuOK in t-BuOH under reflux resulted in a mixture of different compounds. Eventually, the use of 1.5 equivalents of t-BuOK in THF and heating under microwave irradiation for ten minutes at 120 °C selectively provided 3-ethylideneazetidines 10a,b in excellent yields (Scheme 3).¹³

The reactivity study of 3-ethylideneazetidines was expected to be a quite challenging task bearing in mind the stericity hindered and poorly reactive double bond. Prior to evaluating the intrinsic reactivity of this olefinic moiety, the propensity of the azetidine ring to undergo ring opening was investigated. Due to the presence of an electron-donating alkyl group at nitrogen, activation of the azetidine ring toward an azetidinium species is necessary to effect ring-opening processes. N-Acetylation of alkylideneazetidine 10a with 1.5 equivalents of acetyl chloride in CH₂Cl₂ and subsequent ring opening by the displaced chloride ion afforded a mixture of (E)- and (Z)-allylamines 12 (E = MeCO, X = Cl, E/Z = 1:1) after 15 hours under reflux (Table 1, Scheme 3). In a similar manner, the reaction of 10a with one equivalent of benzyl bromide in MeCN gave the corresponding allylamines 13 (Table 1, E = Bn, X = Br, E/Z = 3:2 or vice versa) after 15 hours under reflux. These reactions were straightforward and resulted in a complete conversion of the starting material, although

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**Scheme 1**

**Scheme 2**

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chromatographically inseparable E/Z-mixtures were obtained. In addition, treatment of azetidines 10a,b with 1.5 equivalents of methyl chloroformate in MeCN for 15 hours under reflux resulted in an E/Z-mixture of 4-amino-but-2-enes 14 (E = COOMe, X = Cl, E/Z = 1:1). Upon heating this mixture under microwave irradiation (140 °C, 30 min, 150 W) in DMF, the corresponding new cyclic carbamates 16 were obtained through 6-exo-tet cyclization. These cyclic carbamates can be regarded as interesting compounds with a variety of applications, most notably as precursors for γ-amino alcohols,13 as chiral auxiliaries,16 and as the core substructure in a number of biologically active compounds.17 When azetidine 10a was added to a mixture of 1.3 equivalents of benzyl- or methoxycarbonyl chloride and three equivalents of Et3N in CH2Cl2 and stirred at room temperature for 15 hours in an attempt to effect cycloaddition, the corresponding ring-opened amides 15a,b (E = MeOCH2CO or E = BnOCH2CO, X = Cl, E/Z = 1:1) were formed instead. Apparently, the initial attack of the nucleophilic nitrogen atom across the ketene formed in situ and subsequent ring opening of the azetidine moiety prevailed over the desired cycloaddition reaction.

The reactivity of 3-ethylideneazetidines 10 with respect to electrophilic additions across the exocyclic double bond was evaluated in the next phase of this work. Attempts to prepare halohydrins by treatment of azetidine 10a with one equivalent of NBS in water/THF (1:1) for ten minutes to two days proceeded sluggishly and gave complex mixtures. The outcome of this reaction was shown to be difficult to control and also dependent on the purity of NBS. On the other hand, selective access to the functionalized dibrominated azetidine 17 was achieved by the reaction of azetidine 10a with two equivalents of NBS in CHCl3 or CH2Cl2 under reflux for 15–24 hours. Apparently, the small amount of bromine, released from NBS, was able to react with azetidine 10a to afford 3-bromo-3-(1-bromoethyl)azetidine 17 (X = Br), although in variable yields (30–70%) depending upon the purity of the NBS (Scheme 4). In another approach, the azetidine nitrogen atom was protonated by introducing gaseous HCl to the solution of azetidine 10a in CH2Cl2 for ten minutes, after which one equivalent of mCPBA was added. Instead of the expected spirocyclic azetidinyl epoxide 20a, 3-chloro-3-(1-chloroethyl)azetidine 18 (X = Cl) was obtained in 92% yield (Scheme 4), probably as the result of the electrophilic addition of in situ formed Cl2 to the double bond.14 The vicinal dihalogenated azetidines 17 and 18 were subsequently subjected to reactions with benzylamine or KCN in MeCN in the presence of a catalytic amount of Ag2CO3 or NaI. Unfortunately, these reactions resulted in

### Table 1 Activation and Ring Opening of 3-Ethylideneazetidines 10a,b toward Functionalized Allylamines

<table>
<thead>
<tr>
<th>Substrate</th>
<th>Reaction conditions</th>
<th>E</th>
<th>X</th>
<th>Product (yield, E/Z)</th>
</tr>
</thead>
<tbody>
<tr>
<td>10a</td>
<td>AcCl (1.5 equiv), CH2Cl2, Δ, 15 h</td>
<td>MeCO</td>
<td>Cl</td>
<td>12 (R = H, 100%, E/Z = 1:1)</td>
</tr>
<tr>
<td>10a</td>
<td>BnBr (1 equiv), MeCN, Δ, 15 h</td>
<td>Bn</td>
<td>Br</td>
<td>13 (R = H, 100%, E/Z = 3:2 or vice versa)</td>
</tr>
<tr>
<td>10a</td>
<td>CICOOMe (1.5 equiv), MeCN, Δ, 15 h</td>
<td>COOMe</td>
<td>Cl</td>
<td>14a (R = H, 100%, E/Z = 1:1)</td>
</tr>
<tr>
<td>10b</td>
<td>CICOOMe (1.5 equiv), MeCN, Δ, 15 h</td>
<td>COOMe</td>
<td>Cl</td>
<td>14b (R = Me, 100%, E/Z = 1:1)</td>
</tr>
<tr>
<td>10a</td>
<td>MeOCH2COCl (1.3 equiv) E3N (3 equiv), CH2Cl2, r.t., 15 h</td>
<td>MeOCH2CO</td>
<td>Cl</td>
<td>15a (R = H, 78%, E/Z = 1:1)</td>
</tr>
<tr>
<td>10a</td>
<td>BnOCH2COCl (1.3 equiv) E3N (3 equiv), CH2Cl2, r.t., 15 h</td>
<td>BnOCH2CO</td>
<td>Cl</td>
<td>15b (R = H, 75%, E/Z = 1:1)</td>
</tr>
</tbody>
</table>
the recovery of the starting materials or gave complex mixtures.

In a final attempt to produce the synthetically challenging spirocyclic azetidinyl epoxides 20, the azetidine nitrogen atom in structures 10 was protected by addition of one equivalent of p-TsOH in CH₂Cl₂. Subsequent addition of 1.5 equivalents of mCPBA and heating under reflux for 15 hours afforded highly unstable azetidin-3-ols 19a,b. However, immediate treatment of these alcohols with one equivalent of NaH in THF for 15 hours at room temperature provided the target 1-oxa-5-azaspiro[2.3]hex-3-ene 17 (70%, X = Br) or the corresponding 17 (92%, X = Cl) in excellent yields (Scheme 4). 19 These novel strained spirocyclic systems showed a considerable stability as they could be purified by means of column chromatography on basic Al₂O₃ to provide analytically pure samples. The synthesis of the spiroazetidinyl epoxide moiety has received only very limited attention in the literature. 20 However, these compounds have been shown to be useful intermediates for the preparation of different biologically active molecules. 21 In general, the synthesis and reactivity of different azaspirocyclic scaffolds represent a challenging task for organic chemists and have lately been the subject of significant interest. 20a,22

Bearing in mind that a number of azaspirocycles containing an azetidine moiety can be considered as structural surrogates of commonly employed saturated heterocycles with beneficial inherent structural features, further efforts were devoted to expand the family of novel spiroazetidine building blocks. By analogy with the epoxidation of azetidines 10, the direct aziridination of the double bond could provide an access to novel spirocyclic 1,5-diazaspiro[2.3]hexanes, 23 although the treatment of azetidine 10a with NBS and Chloramine-T as nitrone precursor 24 in MeCN afforded only small amounts of 3-bromo-3-(1-bromoethyl)azetidine 17 and no traces of the corresponding spiro compounds. Using phenyltrimethylammonium tribromide (PTAB) and chloramine- T in MeCN, a complex mixture was also obtained. 25 An alternative route to the synthesis of the spiro-fused aziridinyl azetidine core structure could comprise the ring opening of epoxides 20 with an appropriate amine (i-PrNH₂) in the presence of BF₃·OEt₂, followed by subsequent ring closure of the resulting amino alcohols under Mitsunobu conditions. Although the epoxide ring opening was shown to be successful, the drawback of this procedure involved the very low stability of the β-amino alcohol thus obtained, which underwent immediate decomposition. On the other hand, ring opening of epoxides 20 with three equivalents of NaN₃ and two equivalents of NH₄Cl in acetone/water (8:1) did afford the corresponding azide 21 after 15 hours under reflux (Scheme 4). However, the subsequent ring closure of 21 utilizing Ph₃P in THF gave only complex reaction mixtures. In addition, dihydroxylation of the double bond in azetidines 10a,b with 1.1 equivalents of N-methylmorpholine-N-oxide (NMO) and OsO₄ (5 mol%) in acetone/water (4:1) for four hours at room temperature, followed by an aqueous workup, furnished dihydroxazetidines 22a,b in good yields (Scheme 4). In order to provide an entry to a different class of spirocyclic building blocks, azetidine 22b was treated with 1.1 equivalents of p-TsOH and five equivalents of CuSO₄ in acetone to afford the corresponding novel 5,7-dioxa-2-azaspiro[3.4]octane 23 after stirring under reflux for one day (Scheme 4). This spirocyclic core structure has been found to be present in a number of spiro lactams, suitable for further chemical transformations. 26

In conclusion, 3-bromo-3-ethylazetidines have been shown to undergo ready nucleophilic substitution with

Scheme 4
different nucleophiles, providing a convenient method for the preparation of new 3-alkoxy-, 3-aryloxy-, 3-acetoxy-, 3-hydroxy-, 3-cyano-, 3-carbamoyl- and 3-amino-3-ethylyazetidines. Furthermore, 3-bromo-3-ethylyazetidines can be used as suitable substrates for the preparation of 3-ethylylideneazetidines which, in spite of the presence of a rather inactive double bond, were shown to represent valuable compounds for the preparation of novel functionalized azetidines and spirocyclic azetidine building blocks.

References and Notes

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(13) 3-Ethylidene-1-(4-methylbenzyl)azetidine (10b): To an ice-cooled solution of 3-bromo-3-ethyl-1-(4-methylbenzyl)azetidine (3b) (1.34 g, 5 mmol) in anhyd THF (30 mL), t-BuOK (0.84 g, 1.5 equiv) was added and the mixture was subjected to microwave heating (150 W) for 10 min at 120 °C. Afterwards, the reaction mixture was cooled to rt, filtered, poured into H2O (20 mL) and extracted with EtO (3 × 20 mL). The combined organic extracts were washed with H2O (2 × 15 mL) and brine (20 mL). Drying (MgSO4), filtration of the drying agent and evaporation of the solvent in vacuo afforded azetidine (10b) (0.88 g, 94%), which was purified by silica gel column chromatography to obtain an analytically pure sample; pale yellow oil; Rf 0.22 (petroleum ether–EtOAc, 4:1); yield: 94%. 1H NMR (300 MHz, CDCl3): δ = 1.46–1.51 (m, 3 H), 2.33 (s, 3 H), 3.68 (s, 2 H), 3.79–3.82 (m, 4 H), 5.16–5.24 (m, 1 H), 7.11–7.14, 7.18–7.21 (m, 4 H). IR (neat): 2971, 2805, 1514, 1439, 1358, 1273, 1176, 1042, 1021, 806, 780, 753 cm−1. MS: m/z (%) = 188 (100) [M+ 1]. HRMS (ESI): m/z [M + H]+ C26H26N+ H2O C26H25N+ found: 188.1434; 188.1435.


(19) 5-Benzyl-2-methyl-5-aza-1-oxaspiro[2.3]hexane (20a): To an ice-cooled solution of 1-benzyl-3-hydroxy-3-(1-tosyl-oxyethyl)azetidine (19a; 0.19 g, 0.5 mmol) in anhyd THF (15 mL), NaH (60% suspension; 0.02 g, 1 equiv) was slowly added and the mixture was stirred for 15 h at r.t. The reaction mixture was poured into H2O (20 mL) and extracted with Et2O (3 × 20 mL). The combined organic extracts were washed with H2O (2 × 15 mL) and brine (20 mL). Drying (MgSO4), filtration of the drying agent and evaporation of the solvent in vacuo afforded spirocycle 20a (0.10 g, 95%), which was purified by means of column chromatography on basic alumina in order to obtain an analytically pure sample; pale yellow oil; Rf 0.22 (petroleum ether–EtOAc, 4:1); yield: 95%. 1H NMR (300 MHz, CDCl3): δ = 1.23 (d, J = 5.5 Hz, 3 H), 3.07 (q, J = 5.5 Hz, 1 H), 3.35–3.38, 3.43–3.46 (2 × m, 2 H), 3.60–3.66 (m, 2 H), 3.76 (s, 2 H), 7.23–7.36 (m, 5 H). 13C NMR (75 MHz, CDCl3): δ = 15.5, 56.2, 59.5, 60.4, 61.6, 64.1, 127.3, 128.6, 138.2. IR (neat): 2925, 2831, 1495, 1453, 1363, 1161, 826, 725, 697 cm–1. MS: m/z (%) = 190 (100) [M+ + 1]. HRMS (ESI): m/z [M + H]+ calcd for C12H15NO: 190.1232; found: 190.1232.


