Lewis Acid Catalyzed Rearrangement of Furylcarbinols: The Aza- and Oxa-Piancatelli Cascade Reaction

Leoni I. Palmer, Javier Read de Alaniz*

Department of Chemistry and Biochemistry, University of California-Santa Barbara, Santa Barbara, CA 93106-9510, USA
Fax +1(805)8934120; E-mail: javier@chem.ucsb.edu
Received: 30.08.2013; Accepted: 24.09.2013

Abstract: The acid-catalyzed rearrangement of furylcarbinols is utilized to access 4,5-substituted cyclopentenones. This cascade transformation has currently progressed through an intramolecular rearrangement to the use of alcohols as the nucleophile. Key words: lanthanides, cascade, rearrangement, cyclopentenones, 4π electrocyclization.

The replacement of a multistep chemical synthesis with a cascade reaction is a powerful strategy to improve the efficiency and sustainability of synthesis.1 The potential of a cascade transformation is maximized when combined with straightforward reaction conditions and the formation of widely utilized synthetic building blocks. Incorporating pericyclic reactions within cascade transformations is an interesting subset of this research area; here the benefits are combined with the ability to create new carbon–carbon bonds stereospecifically by simple orbital reorganization.

While a number of important methodologies involving cascade reactions have been developed, we saw the work of Piancatelli and others on the acid-catalyzed rearrangement of furylcarbinols and furfural to be an amazing transformation with, as we felt, room for further expansion in developing its synthetic utility. Herein, we describe the evolution and most efficient strategies employed for these cascade rearrangements and highlight our recent contributions to this area.

In 1976, Piancatelli and co-workers developed the synthesis of 4-hydroxycyclopentenone derivatives by an acid-catalyzed rearrangement of 2-furylcarbinols (Scheme 1, b).2 Despite being discovered over 35 years ago, the Piancatelli reaction still provides the most direct synthetic route to 4-hydroxycyclopentenones and has been the foundation for the synthesis of numerous prostaglandins and other biologically active molecules.3 Recent advances in the Piancatelli rearrangement include the development of microwave-assisted conversions of furfural, which dramatically increases the rate and often leads to high yields.4

**Piancatelli proposed a mechanistic hypothesis that resembles closely the Nazarov reaction (Scheme 1, a).**5 In a Nazarov reaction, a divinyl ketone (1) is transformed to a cyclopentenone (2) in acidic media via a concerted ring closure of a pentadienyl cation (A) to a hydroxycyclopentenyldication (B). In the Piancatelli rearrangement (Scheme 1, b), in the presence of acidic media the furylcarbinol (3) forms a highly reactive oxocarbenium ion, which upon attack by a suitable nucleophile, in this case water, and subsequent ring opening forms a cyclopentadienyl cation (C). Analogous to the Nazarov reaction, this cyclopentadienyl cation can proceed through a 4π-electrocyclic ring closure to give the desired cyclopentenone product (4). De Lera has studied this transformation computationally and determined that the rearrangement probably proceeds through the proposed concerted electrocyclic process and not through an alternative mechanism like an intramolecular aldol reaction or a Prins reaction.6 A year after Piancatelli’s discovery, studies by Lewis and co-workers demonstrated that 4,5-diaminocyclopentenones (5) could be formed from furfural through a related cascade rearrangement, albeit in moderate yield.7 In 2007,
Li and Batey rendered the cascade condensation/ring-opening/electrocyclization process practical and synthetically useful by using lanthanide(III) catalysis (Scheme 2).8

Our motivation for the development of the aza-Piancatelli rearrangement resulted from a literature survey that revealed a lack of methods available for the direct synthesis of 4-aminocyclopentenones. All prior reported methods require multiple steps and typically lack substitution at the adjacent 5-position.9 We hypothesized that an efficient, catalytic aza-Piancatelli rearrangement could serve as a powerful method to streamline the synthesis of biological and medicinal molecules, specifically those bearing nitrogen functionality.

Our investigations began, encouraged by Batey’s use of rare-earth Lewis acids to access bis 4,5-diaminocyclopentenones, with the use of Dy(OTf)$_3$ in catalytic amounts (5 mol%) to facilitate the rearrangement of furylcarbinols with aniline nucleophiles (Scheme 3). To our delight, the rearrangement was efficient, trans-selective and amenable to a wide range of substitutions both on the aniline and on the furylcarbinol.10 We were most encouraged by this initial study, that demonstrates that electron-rich, electron-poor, sterically bulky and even secondary anilines (16) participate successfully in the rearrangement (Scheme 3). Subsequent to our initial report, Reddy et al. and Wang et al. independently demonstrated that phosphomolybdic acid (PMA), In(OTf)$_3$ and La(OTf)$_3$ could also be used to catalyze the aza-Piancatelli reaction.11

Desiring to progress the newly developed aza-Piancatelli rearrangement, a natural advance was to investigate the use of intramolecular substrates (Scheme 4), which met with an equally satisfactory level of success. Methods to construct azaspirocycles usually approach the two main challenges of their synthesis in discrete steps; construction of the 3° carbon center bearing a nitrogen atom and...
the formation of the spirocyclic ring system.\textsuperscript{12} Here, we were most pleased to discover that we can combine these into a single transformation.\textsuperscript{13} Utilizing our previously determined conditions, we saw again that variation of substituents on both the aniline and furylcarbinol portions were well tolerated. Noteworthy was the inclusion of examples that had no substituent on the furylcarbinol (27 and 28; Scheme 4).

We saw fit to apply our cascade rearrangement methodology for the synthesis of biological and medicinal molecules. A structure–activity relationship (SAR) study by Merck revealed 1,2-trans,2,3-trans-cyclopentane-based scaffolds (30) to be comparable to FDA-approved hNK1 antagonist Aprepitant, used as an antiemetic for chemotherapy-induced nausea and vomiting.\textsuperscript{14} We demonstrated that only four steps from our cyclopentenone product (29), formed in 60\% yield on gram scale; Scheme 5) gave access to one such analogue in good yield with the option to easily access other derivatives if desired (30).

\begin{center}
\textbf{Scheme 5} Synthesis of an hNK\textsubscript{1} antagonist
\end{center}

Over the course of these studies it became apparent to us that we were in a position to evaluate the mechanistic possibility that spirocyclic ethers could be a precursor to the more thermodynamically stable oxabicyclic cyclopentenones.\textsuperscript{17c,4} We decided to further investigate this relationship and were delighted to find that oxaspirocyclic cyclopentenone framework could be used to access fused oxabicycles (Scheme 8).\textsuperscript{18} Treatment of oxaspirocycles (43) with Amberlyst\textregistered{} 15 in heated toluene resulted in clean conversion to fused cyclopentenones (44). This is a general transformation utilizing an acid-facilitated skeletal rearrangement to the thermodynamic product.

In summary, the Piancatelli rearrangement has been extended to include nitrogen nucleophiles and other oxygen nucleophiles in the synthesis of highly stereoselective and variably substituted cyclopentenones. We have demonstrated a transformation of value to the synthetic community for use in the construction of cyclopentenones,
Scheme 8 Conversion of spirocycles into fused bicycles

spirocycles and fused bicyclic frameworks and promises to deliver further exciting developments in the future.

Acknowledgment

This work was supported by UCSB and NSF (CAREER Award CHE-1057180). Additional support was kindly provided by Eli Lilly (New Faculty Award to J.R.A.).

References and Notes


(15) Previously undisclosed results.

