SYNLETT **Spotlight 448**

This feature focuses on a reagent chosen by a postgraduate, highlighting the uses and preparation of the reagent in current research

N,N,N',N'-Tetramethylethane-1,2diamine (TMEDA)

Compiled by Bharat Kumar Allam

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Dedicated to my parents and my esteemed supervisor Professor Dr. Krishna Nand Singh

Introduction

N, N, N', N'-Tetramethylethane-1,2-diamine (TMEDA) is a colorless, liquid, organic base with a fishy odor. TMEDA serves as an excellent donor of methylene and formyl groups under base switching.¹ It functions as an important ligand and additive for a broad spectrum of transitionmetal-catalyzed reactions because of its unique ability to enhance the rate of metalation for a variety of compounds.² Because of its easy solvating and bidentate ligand properties it stabilizes and activates a number of organometallic reagents and metal salts. TMEDA exhibits a special affinity towards lithium ions and this feature allows BuLi/TMEDA to metalate many substrates.³

Abstracts

(A) (Phenylsulfonyl)difluoromethylation 4-i-Pr-CoH PhO₂SF CuE₂:2H₂O (20 mol%) Hu's group has developed an efficient protocol for vinylic difluoro-TMEDA (25 mol%) methylation of α , β -unsaturated carboxylic acids catalyzed by H₂O-DCE, 80 °C, 12 h CuF₂·2H₂O and TMEDA in H₂O-DCE.⁴ CF₂SO₂Ph 91% yield (B) Direct ortho Arylation of Electron-Deficient N-Heteroarenes Da and co-workers reported an efficient metal-free protocol for the 1. TMEDA, PhMe, 50 °C ArMaB direct ortho arylation of electron-deficient N-heteroarenes with aro-2. air, 12 h matic Grignard reagents assisted by TMEDA. Differently substitut-62-93% vield ed quinolines, isoquinolines, pyridines, and quinoxaline have participated well in the addition reactions with Grignard reagents.5 (C) Syntheses of Unsymmetrical Conjugated Diynes The aerobic oxidative coupling of two different terminal alkynes has OMe NiCl₂·6H₂O, Cul been achieved through the cooperative catalysis of NiCl₂·6H₂O and TMEDA (20 mol%) ć CuI. The protocol utilizes 20 mol% TMEDA as the ligand and air or Et₃N, THF, r.t., air, 20 h O_2 as the oxidant.⁶ 73% vield (D) Cyanation of Aryl Halides Shen et al. have discovered an inexpensive and non-toxic cyanation Pd(OAc)₂ of electron-rich and electron-deficient aryl halides using ethyl cya-DPPE, TMEDA, Na₂CO₃ noacetate as a source of the cyano group. The reaction has been con-DMF ducted using Pd(OAc)₂ as a catalyst and in the presence of 1,2-

99% yield

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TMEDA in DMF.

bis(diphenylphosphino)ethane (DPPE), potassium iodide, and

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