This feature focuses on a reagent chosen by a postgraduate, highlighting the uses and preparation of the reagent in current research.

**Ytterbium Trifluoromethansulfonate**

Compiled by Tran Anh Tuan

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**Introduction**

Ytterbium trifluoromethansulfonate \([\text{Yb(OTf)}_3]\) has been widely used in organic syntheses in the last few years.\(^1\) \(\text{Yb(OTf)}_3\) is a strong Lewis acid\(^2\) due to the hard character of \(\text{Yb}^{3+}\) ion and the presence of electron-deficient triflate in its coordination sphere. In contrast to traditional Lewis acids, such as \(\text{AlCl}_3\), \(\text{BF}_3\), \(\text{TiCl}_4\), and \(\text{SnCl}_4\), which are often used in stoichiometric amounts, only catalytic amounts of \(\text{Yb(OTf)}_3\) are necessary. Moreover it can be easily recovered and reused without loss of activity. Interestingly, \(\text{Yb(OTf)}_3\) remains catalytically active in the presence of many Lewis bases containing nitrogen, oxygen, phosphorus or sulfur atoms. The resulting water-compatibility of \(\text{Yb(OTf)}_3\)\(^3\) is one of its well-known advantages, with respect to traditional Lewis acids that are very sensitive and easily decomposed or deactivated in the presence of small amounts of water. The most interesting point from a synthetic point of view is that \(\text{Yb(OTf)}_3\)-catalyzed reactions are clean, while \(\text{Yb(OTf)}_3\) is regarded as environmentally friendly catalyst. Ytterbium triflate is prepared by heating ytterbium(III) oxide or chloride in an aqueous trifluoromethansulfonic acid solution (Scheme 1).\(^4\)\(^-\)\(^5\)

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\begin{align*}
\text{Yb}_2\text{O}_3 + 6 \text{TfOH} & \rightarrow 2 \text{Yb(OTf)}_3 + 3 \text{H}_2\text{O} \\
\text{YbCl}_3 + 3 \text{TfOH} & \rightarrow \text{Yb(OTf)}_3 + 3 \text{HCl}
\end{align*}
\]

**Scheme 1**

This reagent has been used in numerous organic transformations,\(^1\) e.g. in aldol reactions,\(^6\) Kharasch-type additions,\(^7\) glycosylations,\(^8\) Friedel–Crafts acylations,\(^9\) dealkoxyacetylations,\(^10\) syntheses of β-enaminones,\(^11\) etc.

This article describes some major applications in organic synthesis in the recent years.

**Abstracts**

**(A) Friedel–Crafts Acylation:**
The acylation of 1-methylpyrrole has been reported recently.\(^1\)\(^2\) The reaction is carried out in [bpy][BF_4] with a catalytic amount of \(\text{Yb(OTf)}_3\) (10 mol%) at room temperature. Good yields were obtained (80–93%), but the reaction fails without catalyst. Moreover, the catalyst can be recycled three times without loss of activity.

**(B) Crotonation:**
3-Acryl acrylic acids possess a high potential in the synthesis of biologically or pharmaceutically active compounds, such as 4,\(^13\) 5,\(^14\) and 6.\(^15\) Recently, Gorobets and co-workers described a new facile protocol for the synthesis of aromatic and heteroaromatic 3-acryl acrylic acids 9 in good yields (54–78%).\(^16\) Aromatic ketones 7, glyoxylic acid monohydrate 8, and \(\text{Yb(OTf)}_3\) (2.5 mol%) are reacted under microwave irradiation.
(C) Tosylation:
Most tosylations use triethylamine or pyridine as a base in the reaction of appropriate alcohols with the tosylating agents. In 2004, Schirrmacher and Comagic reported the low-yielding tosylation of using TsCl and pyridine. Gratifyingly, when Yb(OTf)3 is used, the tosylation with Ts2O proceeded in excellent yield (85%). These conditions were also applied for primary and secondary alcohols and provided the tosylates in good yields (75–89%).

(D) TEMPO-Mediated Oxidation:
Vatèle described a new method for the oxidation of alcohols with iso-dibenzylether relying on the utilization of the TEMPO/PhIO system as oxidizing source. However, when 4-phenyl-butan-1-ol was treated with PhIO and TEMPO, only 5% of 4-phenyl-butan-1-ol was obtained. In the presence of Yb(OTf)3 (2 mol%), the expected aldehyde was obtained in good yield (83%). The triflate can also catalyze the oxidation of several primary or secondary alcohols into the corresponding aldehydes or ketones in good to excellent yields (70–94%).

(E) One-Pot Multicomponent Synthesis of Substituted Imidazoles:
Yb(OTf)3 has been used for the synthesis of substituted imidazoles through three-component condensation of benzil, other sugar peracetates, such as 12, aldehydes or ketones in good to excellent yields (76–94%).

(F) Selective Anomeric Deacetylation:
Selective anomeric deacetylation is a key step in the oligosaccharide synthesis. Yb(OTf)3 can promote the selective anomeric deacetylation of compound 14 (an important synthon involved in the synthesis of heparin sulphate fragments). The reaction is carried out using catalytic amounts of Yb(OTf)3 (5 mol%) and gave compound 15 in good yield (75%). However, Nd(OTf)3 proved to be a superior catalyst. This protocol also gave good yields (61–85%) when applied to other sugar peracetics, such as α- or β-D-glucopyranose or α-D-xylpyranose peracetates. One of the most striking features is that Yb3+ and Nd3+ catalyzed the transesterification of the anomeric acetate without catalyzing the methyl glycoside formation.

References