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This feature focuses on a reagent chosen by a postgraduate, highlighting the uses and preparation of the reagent in current research

Simmons–Smith Reagent (Et₂Zn, CH₂I₂): An Efficient Reagent in Organic Synthesis

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Introduction

The diethylzinc/diiodomethane or zinc-diiodomethane (Et_2Zn , CH_2I_2 or Zn, $CH_2I_2 = ICH_2ZnI$) known as Simmons–Smith reagent is probably the best known carbenoid reagent in organic syntheses.¹ This ether soluble reagent has been used mainly for the conversion of alkenes into cyclopropanes¹ via stereospecific and supra-

facial CH₂ addition using chiral auxiliaries,² reagents³ and catalysts.⁴

Due to the presence of cyclopropanes in many biologically and medicinally important molecules,⁵ natural products,⁶ essential oils⁷ and the marine cyanobacteriums,⁸ some recent applications of Simmons–Smith reagent are reported herein.

Abstracts

(A) Asymmetric reaction of allylic alcohols with Al Lewis acid/N Lewis base bifunctional Al(salalen) catalyst in the presence of Simmons–Smith reagent has been reported by Katsuki and Shitama in quantitative yields with high enantiomeric excess. The hydroxyl group is a prerequisite serving as an anchor for zinc.⁹

(B) A catalytic asymmetric Simmons–Smith cyclopropanation of silyl enol ethers 1 using dipeptide 2 as a ligand has been described. A variety of optically active cyclopropyl silyl ethers 3 can be obtained in high yields and with an ee up to 96%. The dipeptide can be recovered after the reaction in good yield and reused without the loss of reactivity or enantioselectivity.¹⁰

(C) White et al. have reported asymmetric total syntheses of solandelactones. The key step in these syntheses involved directed Simmons–Smith cyclopropanation which this reaction preceded in quantitative yield (97%).¹¹

(D) A highly diastereoselective cyclopropanation protocol for allylic amines 1 has been demonstrated by Davies et al. giving access to complete conversion into *syn-2* in 98% de with Simmons–Smith reagent, and *anti-3* in 98% de in the presence of TFA.¹²

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(E) Recently, an efficient asymmetric synthesis of conformationally constrained (2*S*,3*S*)-piperidinedicarboxylic acid derivatives has been reported by Zhuo et al. This was the first successful example of cyclopropanation of an allylic amine by masking the amine as a carbamate without the presence of a chelating group (OH or OR) using Simmons–Smith reagent.¹³

(F) A highly stereoselective synthesis of chiral aminocyclopropanes through cyclopropanation of chiral enamides using the Simmons–Smith reagent has been explored by Hsung et al. Various substrates are transformed with moderate to good diastereoselectivity to cyclopropane products. The method includes the synthesis of the cyclopropane fragment of the nucleoside.¹⁴

(G) A series of novel fluorocyclopropyl nucleosides with antiviral activity has been synthesized starting from acetol using the Simmons–Smith reagent. All the synthesized nucleosides were assayed against several viruses.¹⁵

TBDMSO OH
$$\frac{\text{Et}_2\text{Zn, CH}_2\text{l}_2}{\text{CH}_2\text{Cl}_2, 77\%}$$

(H) The Simmons–Smith reagent has been applied to prepare a new class of haloalkylzinc compounds leading to transition metal carbenes. Halomethylzinc and halobenzylzinc compounds react with ruthenium and iridium complexes to form methylene and benzylidene complexes including the Grubbs catalyst. ¹⁶

$$\begin{array}{c} \text{CH}_2 \text{I}_2 + \text{Et}_2 \text{Zn} \xrightarrow{\text{CH}_2 \text{CI}_2} \quad \text{Zn}(\text{CH}_2 \text{I})_2 \xrightarrow{\text{(PPh}_3)_3 \text{RuCI}_2} \\ & -30 \text{ °C} \end{array} \xrightarrow{\text{PCy}_3} \begin{array}{c} \text{CI} \\ \text{Ru=CI} \\ \text{PCy}_3 \end{array}$$

(I) The total synthesis of the natural metabolite (+/-)-cascarillic acid 3 has been achieved by a sequential crossmetathesis and Simmons–Smith cyclopropanation between 1-octene 1 with an appropriate unsaturated carboxylic acid 2. Interestingly, the smooth conditions required for both the cross-metathesis and the cyclopropanation allow the combination of the two processes in one single pot. Compared to other racemic syntheses, this high yielding (up to 90%), selective (the *E/Z* ratio near 80:20) and shorter procedure is expected to be generalized to other naturally occurring *trans*-cyclopropane derivatives.¹⁷

Mes N N Mes

CI., Ru

1) CH₂Cl₂,
$$\Delta$$
, Cl P Cy₃ Ph

+ 2) Et₂Zn, CH₂l₂, CH₂Cl₂

2. CO₂H r.t., 48 h, 90% HO₂C 3

References

- (a) Simmons, H. E.; Smith, R. D. J. Am. Chem. Soc. 1959, 81, 4256. (b) Simmons, H. E.; Smith, R. D. J. Am. Chem. Soc. 1958, 80, 5322. (c) Charette, A. B.; Beauchemin, A. Simmons–Smith Cyclopropanation Reaction In Organic Reactions; John Wiley & Sons: New York, 2001, 58, 1–415.
- (2) Kang, J.; Lim, G. J.; Yoon, S. K.; Kim, M. Y. J. Org. Chem. 1995, 60, 564.
- (3) Charette, A. B.; Juteau, H.; Lebel, H.; Molinaro, C. J. Am. Chem. Soc. 1998, 120, 11943.
- (4) Balsells, J.; Walsh, P. J. J. Org. Chem. 2000, 65, 5005.
- (5) Lebel, H.; Marcoux, J.-F.; Molinaro, C.; Charette, A. B. *Chem. Rev.* **2003**, *103*, 977.
- (6) Wessjohann, L. A.; Brandt, W.; Thiemann, T. Chem. Rev. 2003, 103, 1625.
- (7) (a) Roberts, I. O.; Baird, M. S.; Liu, Y. *Tetrahedron Lett.* 2004, 45, 8685. (b) Cheeseman, M.; Bull, S. D. *Synlett* 2006, 1119.

- (8) Avery, T. D.; Culbert, J. A.; Taylor, D. K. Org. Biomol. Chem. 2006, 4, 323.
- Shitama, H.; Katsuki, T. Angew. Chem. Int. Ed. 2008, 47, 2450.
- (10) Du, H.; Long, J.; Shi, Y. Org. Lett. 2006, 8, 2827.
- (11) White, J. D.; Martin, W. H. C.; Lincoln, C.; Yang, J. Org. Lett. 2007, 9, 3481.
- (12) Davies, S. G.; Ling, K. B.; Roberts, P. M.; Russell, A. J.; Thomson, J. E. *Chem. Commun.* 2007, 4029.
- (13) Zhuo, J.; Burns, D. M.; Zhang, C.; Xu, M.; Weng, L.; Qian, D.-Q.; He, C.; Lin, Q.; Li, Y.-L.; Shi, E.; Agrios, C.; Metcalf, B.; Yao, W. Synlett 2007, 460.
- (14) Song, Z.; Lu, T.; Hsung, R. P.; Al-Rashid, Z. F.; Ko, C.; Tang, Y. Angew. Chem. Int. Ed. 2007, 46, 4069.
- (15) Kim, A.; Hong, J. H. Eur. J. Med. Chem. 2007, 42, 487.
- (16) Poverenov, E.; Milstein, D. Chem. Commun. 2007, 3189.
- (17) Salim, H.; Piva, O. Tetrahedron Lett. 2007, 48, 2059.