SYNLETT Spotlight 278

Iodoxybenzene (PhIO₂)

Compiled by Cai Zhang

This feature focuses on a reagent chosen by a postgraduate, highlighting the uses and preparation of the reagent in current research

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SPOTLIGHT

Introduction

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Iodoxybenzene (PhIO₂, mp 230 °C) is the first organic compound of iodine(V) prepared by Willgerodt 100 years ago. Specifically, it can be prepared by disproportionation of iodosylbenzene under steam distillation. Ten years ago iodoxybenzene has been used for the synthesis of cadalenquinone, a unsaturated carbonyl compounds becompounds, oxodienes, epoxides, fulnine imines, and sulfones. In this Spotlight recent applications using PhIO₂ in organic syntheses are summarized.

Preparation

Iodoxybenzene (PhIO₂) can be directly prepared using commercial 40% peracetic acid or sodium periodate by dissolving iodobenzene in water (Scheme 1).⁴

Scheme 1

Abstracts

(A) Kita et al. show the solubilization and activation of $PhIO_2$ by adding catalytic amounts of both cetyltrimethylammonium bromide (CTAB) or $MgBr_2$ and a chiral tartaric acid derivative. Both reagents were indispensable for the enhancement of chemical and optical yields in the oxidation of sulfides to sulfoxides.⁵

Ar
$$\stackrel{\text{PhIO}_2 \text{ (55 mol\%)}}{\text{R}}$$

$$\frac{\text{MgBr}_2 \text{ (20 mol\%)}}{\text{(+)-dibenzoyl-D-tartaric acid (10 mol\%)}} \stackrel{\text{O}}{\text{Ar}} \stackrel{\text{N}}{\text{S}} \stackrel{\text{R}}{\text{R}}$$

$$43-63\% \text{ ee}$$

$$+20.0 \text{ °C}$$
excellent yields

(B) The application of $PhIO_2$ in the oxidation reaction of phosphorous, phosphorothiono and phosphoroseleno compounds into the corresponding $\equiv P(O)$ analogues was demonstrated by Lopusinski et al.^{6a} The direct conversion of ammonium dialkyl phosphoroselenoates into the dialkyl phosphoramidates using iodoxybenzene has been developed by the same group.^{6b}

$$X = \text{lone electron pair, S, Se}$$

$$R^{1} = \text{ONH}_{3}R^{3} \xrightarrow{\text{PhIO}_{2}} R^{2} \xrightarrow{\text{PhIO}_{2}} R^{2} \xrightarrow{\text{NHR}^{3}} R^{2} \xrightarrow{\text$$

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(C) The reaction performed between N-sulfinylphosphoramidate and $PhIO_2$ in the presence of diethylphosphoramidate results in the formation of N,N-bis(diethylphosphor)sulfamide.⁷

(D) Catalytic allylic oxidation of alkenes into enones and oxidation of aryl alkyl ketones into the corresponding ketoacids with a fluorous seleninic acid as catalyst in conjunction with $PhIO_2$ as reoxidant have been developed by Zou and co-worker.

(E) Recently, air-driven conversion of alcohols into carbonyl compounds in water using catalytic amounts of PhIO₂ has been reported by Liu and co-workers.⁹

$$\begin{array}{c} \text{R}^1 \\ \text{OH} \end{array} \xrightarrow{ \begin{array}{c} \text{PhIO}_2 \text{ (1 mol\%), Br}_2 \text{ (2 mol\%), NaNO}_2 \text{ (1 mol\%)} \\ \text{air, H}_2\text{O, 55 °C} \end{array}} \begin{array}{c} \text{R}^1 \\ \text{R}^2 \\ \text{42-98\%} \end{array}$$

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