SYNLETT Spotlight 267

This feature focuses on a reagent chosen by a postgraduate, highlighting the uses and preparation of the reagent in current research

Trimethylsilylacetonitrile (TMSAN)

Compiled by Penélope Merino-Montiel

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TMSAN

Scheme 1

Introduction

Trimethylsilylacetonitrile (TMSAN) is a popular commercial reagent that has been used in the synthesis of several useful synthetic building blocks. TMSAN is usually used as a nucleophile, mainly in additions to a carbonyl group,¹ in order to obtain cyanomethylated adducts, or in

Abstracts

(A) The condensation of TMSAN with ketones in the presence of the catalytic *t*-Bu-P4 base to give cyanoalkenes has been reported;³ title compounds were prepared in a 63-80% yield. The reaction was also extended to the condensation of TMSAN with *N*-arylform-anilides (ArNRCHO) in the preparation of enamine derivatives.

(B) The addition of TMSAN to the carbonyl group of benzaldehyde catalyzed by KF/Al₂O₃ in THF allows the preparation of 3-phenyl-3-(trimethylsilyloxy)propionitrile as the major product (85%), together with cinnamonitrile (15%). A change of the solvent to DMF leads to the α , β -unsaturated nitrile as main product (61%).⁴

(C) Langer and co-workers reported a one-pot cyclization of epibromohydrin (EBH) with TMSAN in the presence of a strong base and catalyzed by a Lewis acid to form trimethylsilylcyano cyclopropane with excellent diasteroselectivity (E/Z > 98:2) and moderate yield (65%).⁵

(D) Recently, 2-(2-cyanoethyl)aziridines were obtained by treatment of TMSAN with an equimolecular amount of *n*-BuLi followed by the addition of 1-arylmethyl-2-(bromomethyl)-aziridines.⁶ Title compounds were obtained in moderate yield (41–65%).

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 $R^1 = Ph; R^2 = Ph, Me, (E)-CH=CHPh$

TMSAN 1) LDA (2.3 equiv) 2) EBH, LiClO₄ 3) H₂O OH

TMSAN

65% (*E*/*Z* > 98:2)

63-80%

15% (*E*/*Z* = 60:31)

TMS

ΏN





OTMS

85%

nucleophilic substitution reactions, for example in the

preparation of functionalized cyclopropanes.²



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94% (dr = 98:2)

Ar

CO₂Me

CH2TMS

CN

4-t-BuC₆H₄

4-Ph(CO)C₆H₄

4-Me₂NC₆H₄

2-Me₂NC₆H₄

Yield (%)

87

92

83

83

(E) TMSAN has been used as a nucleophile to open the acetal ring in sugar spiroketals. The nucleophilic attack of TMSAN on the anomeric carbon can take place either through the nitrogen atom or through the methylenic carbon⁷ to give a spironucleoside or a spiro-C-glycoside, respectively.



TMSAN

Pd/L (2%), ZnF₂ (0.5 equiv)

TMSAN

Rh2(OAc)4 (1 mol%),

toluene, 60 °C

TMSAN

Ni(cod)₂, LA

DMF, 90 °C

ArBr + TMSAN

CO₂Me

PhONn-Bu₄ (1.0 equiv)

THF, -78 °C

ArCH₂CN

ĊI 70%

(F) There are a few reports of cyanomethylation of aldehydes and ketones using TMSAN as nucleophile. Suto et al. have developed a copper fluoride catalyzed addition to the carbonyl group with high yields.⁸

(G) β -Amino nitriles can be obtained by the condensation between chiral *N*-(*tert*-butylsulfonyl)imines and TMSAN in the presence of a Lewis base with excellent diasteroselectivity.⁹ Optimization of the reaction conditions by changing the Lewis base, temperature, and solvent showed that the best yields and diasteroisomeric ratios were obtained when PhON*n*-Bu₄ was used as a Lewis base in THF at -78 °C.

(H) Wu and Hartwig described the monoarylation of nitriles by means of the coupling of TMSAN with aryl halides in the presence of palladium catalysts and zinc fluoride.¹⁰

(I) A singular reaction of transannulation of pyridotriazoles with TMSAN in the presence of $Rh_2(OAc)_4$ to afford a N-fused imidazopyridine was described by Chuprakov et al.¹¹ Rhodium-catalyzed transannulations of 1-sulfonyl triazoles were recently reported by the same group.¹²

(J) Nakao et al. reported the Ni-catalyzed carbocyanation of 4-octyne with TMSAN in the presence of a Lewis acid (LA) to give a highly substituted allylsilane in modest yield (29%).¹³

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TMSH₂C

Pr[/] 29%

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